

AQA **Surname** _____**Other Names** _____**Centre Number** _____**Candidate Number** _____**Candidate Signature** _____**GCSE****COMBINED SCIENCE: SYNERGY****Higher Tier Paper 1 Life and
environmental sciences****8465/1H****Tuesday 15 May 2018 Afternoon H****Time allowed: 1 hour 45 minutes****At the top of the page, write your surname
and other names, your centre number,
your candidate number and add your
signature.****[Turn over]**

For this paper you must have:

- **a ruler**
- **a scientific calculator**
- **the periodic table (enclosed)**
- **the Physics Equations Sheet (enclosed).**

INSTRUCTIONS

- **Use black ink or black ball-point pen.**
- **Answer ALL questions in the spaces provided. Do not write on blank pages.**
- **Do all rough work in this book. Cross through any work you do not want to be marked.**
- **In all calculations, show clearly how you work out your answer.**



INFORMATION

- **The maximum mark for this paper is 100.**
- **The marks for questions are shown in brackets.**
- **You are expected to use a calculator where appropriate.**
- **You are reminded of the need for good English and clear presentation in your answers.**

DO NOT TURN OVER UNTIL TOLD TO DO SO



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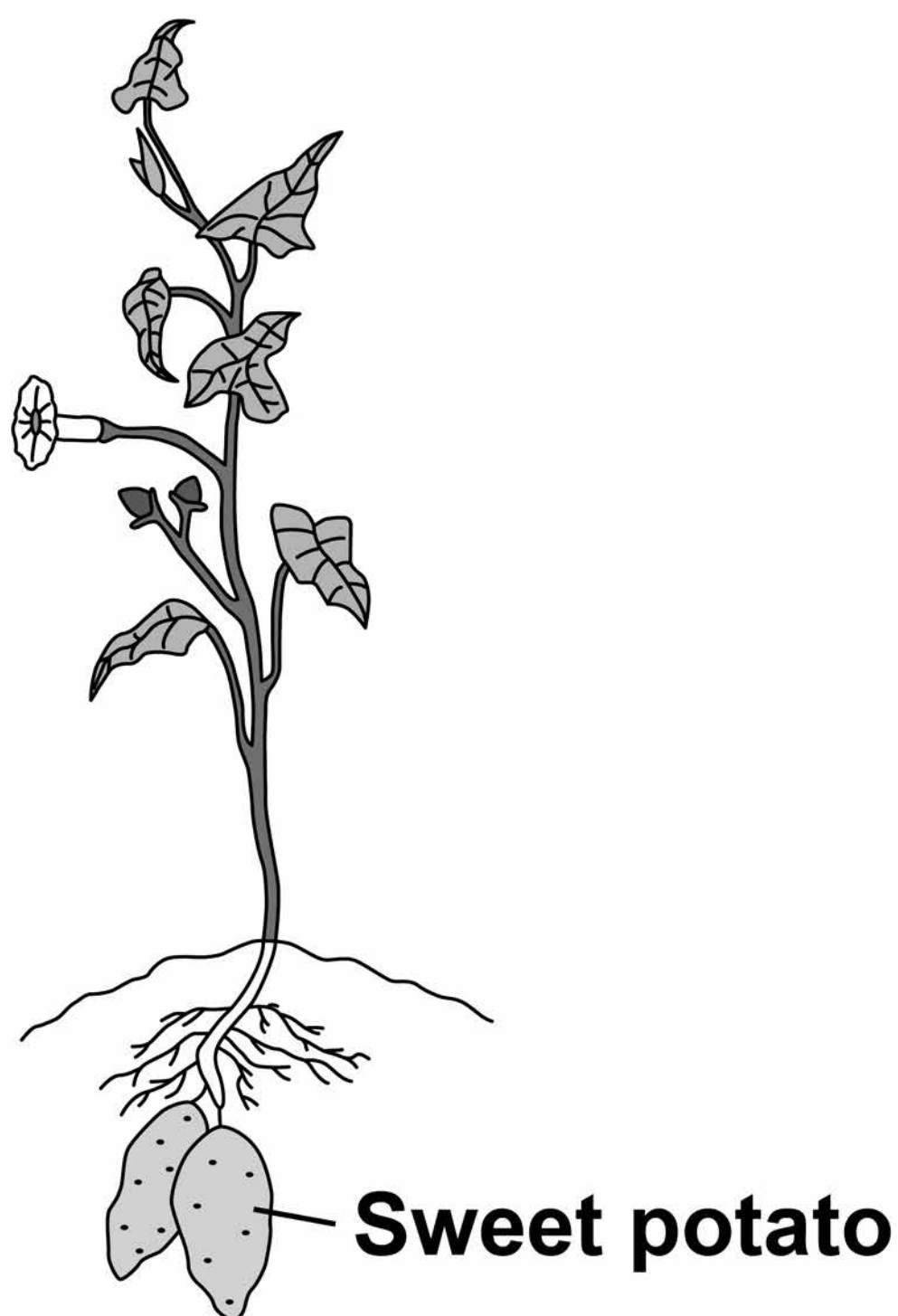


0 1

FIGURE 1 shows a sweet potato plant.

The sweet potatoes grow underground and can be cooked and eaten.

FIGURE 1



[Turn over]



TABLE 1 shows some of the nutrients in cooked sweet potato.

TABLE 1

Nutrient	Mass in grams per 100 grams of cooked sweet potato
Water	73.83
Protein	2.01
Fat	0.15
Total carbohydrate	20.71
of which sugars	6.55
Fibre	3.30



7

0 1 . 1 After cooked sweet potato is digested, sugars (including glucose) pass into the blood.

Give TWO other soluble molecules that would pass into the blood after cooked sweet potato is digested. [2 marks]

1 _____

2 _____

0 1 . 2 Calculate the mass of sugars in 180 g of cooked sweet potato.

Use the information from TABLE 1. [1 mark]

Mass of sugars = _____ g

[Turn over]



[Turn over]

9



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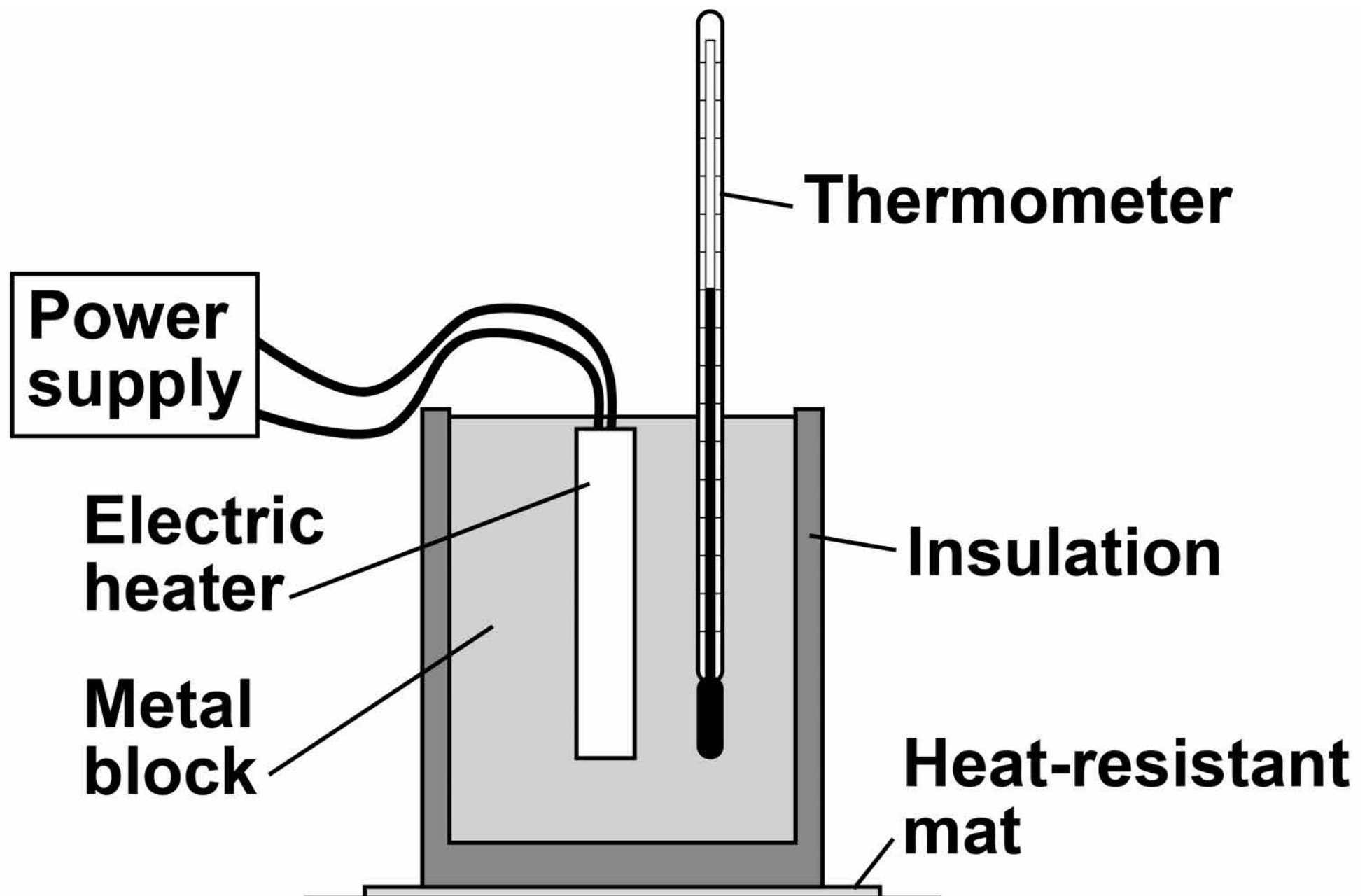


0 2

A student investigated how the temperature of a metal block changed with time.

An electric heater was used to increase the temperature of the block.

The heater was placed in a hole drilled in the block as shown in **FIGURE 2**.

FIGURE 2**[Turn over]**

12

The student measured the temperature of the metal block every 60 seconds.

TABLE 2 shows the student's results.

TABLE 2

Time in s	Temperature in °C
0	20.0
60	24.5
120	29.0
180	31.0
240	31.5

0 2 . 1 Complete the graph of the data from TABLE 2 on FIGURE 3.

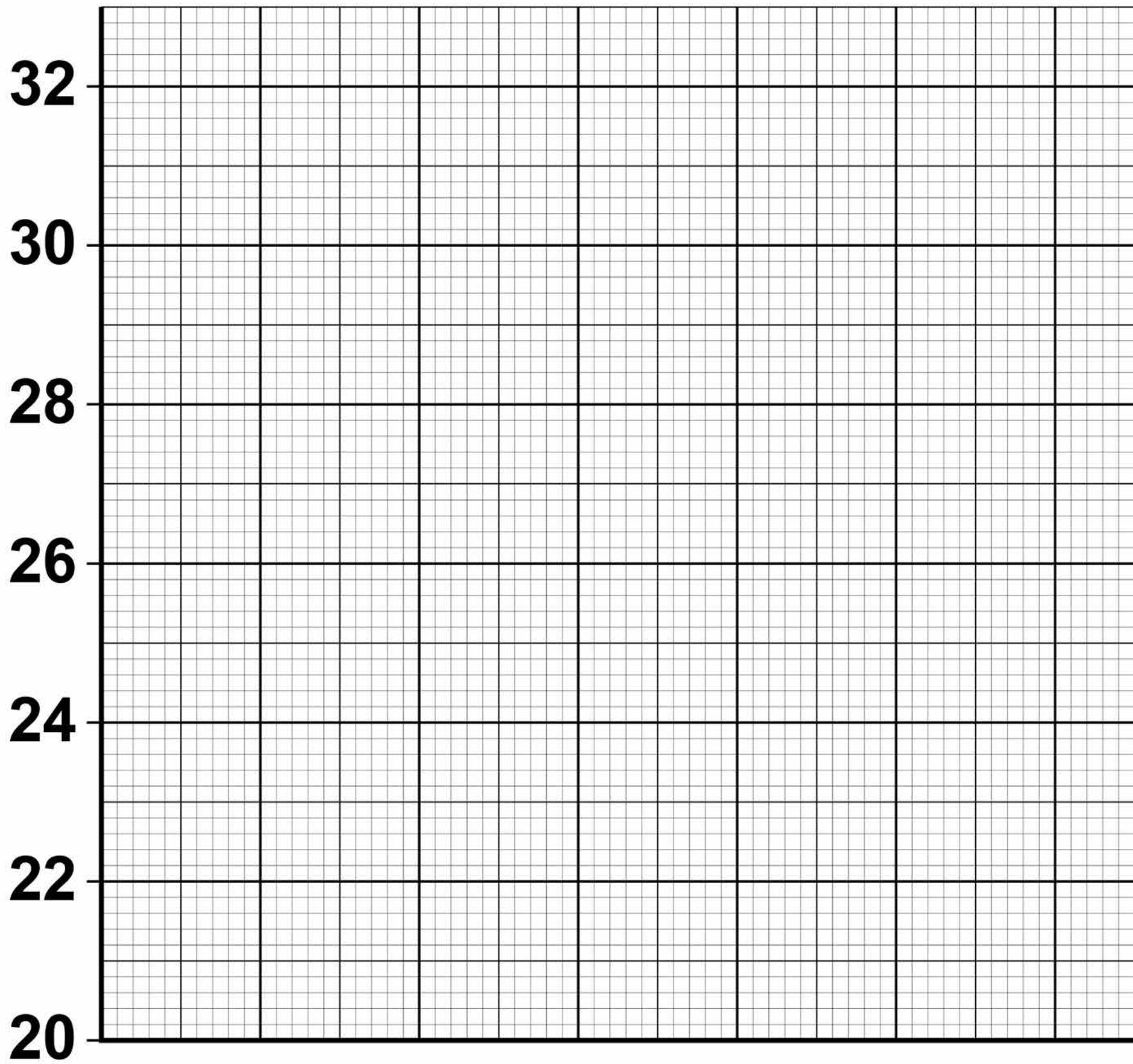
- Choose a suitable scale for the x-axis.
- Label the x-axis.
- Plot the student's results.
- Draw a line of best fit.

[4 marks]



FIGURE 3

Temperature
in °C



[Turn over]



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15

0 2 . 2 The rate of change of temperature of the block is given by the gradient of the graph on page 13.

**Determine the gradient of the graph over the first 60 seconds.
[2 marks]**

Gradient = _____

[Turn over]



16

0 2 . 3 The metal block had a mass of 1.50 kg

The specific heat capacity of the metal was 900 J/kg °C

Calculate the change in thermal energy of the metal during 240 seconds.

Use the Physics Equations Sheet.

Give your answer in kilojoules.
[4 marks]



17

Change in thermal energy =

kJ

[Turn over]



18**Repeat of TABLE 2**

Time in s	Temperature in °C
0	20.0
60	24.5
120	29.0
180	31.0
240	31.5



0 2 . 4 Another student repeated the investigation.

**Give TWO variables this student would need to control to be able to compare their results with the results in TABLE 2 on page 18.
[2 marks]**

1

2

12

[Turn over]



0 3 There are several methods of contraception.

0 3 . 1 Draw ONE line from each method of contraception to how the method works. [2 marks]

**Method of
contraception**

**How the
method works**

diaphragm

**prevents
embryo
implanting**

**intrauterine
device**

**prevents
release of the
egg**

**oral
contraceptive**

**prevents
sperm
reaching the
egg**



0 3 . 2 When a new oral contraceptive is tested on volunteers, the contraceptive is first given at a low dose. Later, the dose is increased.

Why are new drugs given at low doses at first? [1 mark]

[Turn over]

03.3 TABLE 3 shows information about three methods of contraception.

TABLE 3

	Condom	Oral contraceptive	Hormone skin patch
Percentage (%) effectiveness	98.0	99.7	99.8
How contraception is obtained	From shops or sexual health clinic	From doctor or clinic	From doctor or sexual health clinic
Possible side effects	No serious side effects	Headaches, nausea, high blood pressure	Headaches, nausea, blood clots



**Evaluate the use of these contraceptive methods.
[6 marks]**

[Turn over]



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24



9

[Turn over]



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0 4

There is limited evidence about the Earth's early atmosphere because of the age of the Earth.

0 4 . 1

The Earth is 4.6 billion years old.

Which is the correct age of the Earth? [1 mark]

Tick ONE box.

4.6×10^3 years

4.6×10^6 years

4.6×10^9 years

4.6×10^{12} years

[Turn over]



Scientists think that the Earth's early atmosphere may have been similar to the atmosphere on Mars today.

Look at TABLE 4.

TABLE 4

Gas	Concentration of gas in the atmosphere today in parts per million	
	Mars	Earth
Nitrogen	27 000	780 000
Oxygen	1 300	210 000
Argon	16 000	9 300
Carbon dioxide	950 000	400
Carbon monoxide	800	trace



29

0 4 . 2 Calculate the percentage increase in nitrogen from the Earth's early atmosphere to the atmosphere today.

Assume the Earth's early atmosphere was the same as the atmosphere today on Mars.

Give your answer to 2 significant figures. [3 marks]

Percentage increase in nitrogen
= _____ %

[Turn over]



30

0 4 . 3 Which process releases carbon monoxide into the Earth's atmosphere? [1 mark]

Tick ONE box.

Aerobic respiration

Bacterial decomposition

Incomplete combustion

Photosynthesis



31

0 4 . 4 Explain how the oceans were formed in the first billion years of the Earth's existence.
[2 marks]

0 4 . 5 Describe how the increase in greenhouse gases has increased the mass of liquid water in the oceans. [1 mark]

8

[Turn over]



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0	5
---	---

Alpha, beta and gamma are types of nuclear radiation.

0	5	.	1
---	---	---	---

Explain why gamma emission does NOT change the atomic number of an element. [2 marks]

[Turn over]



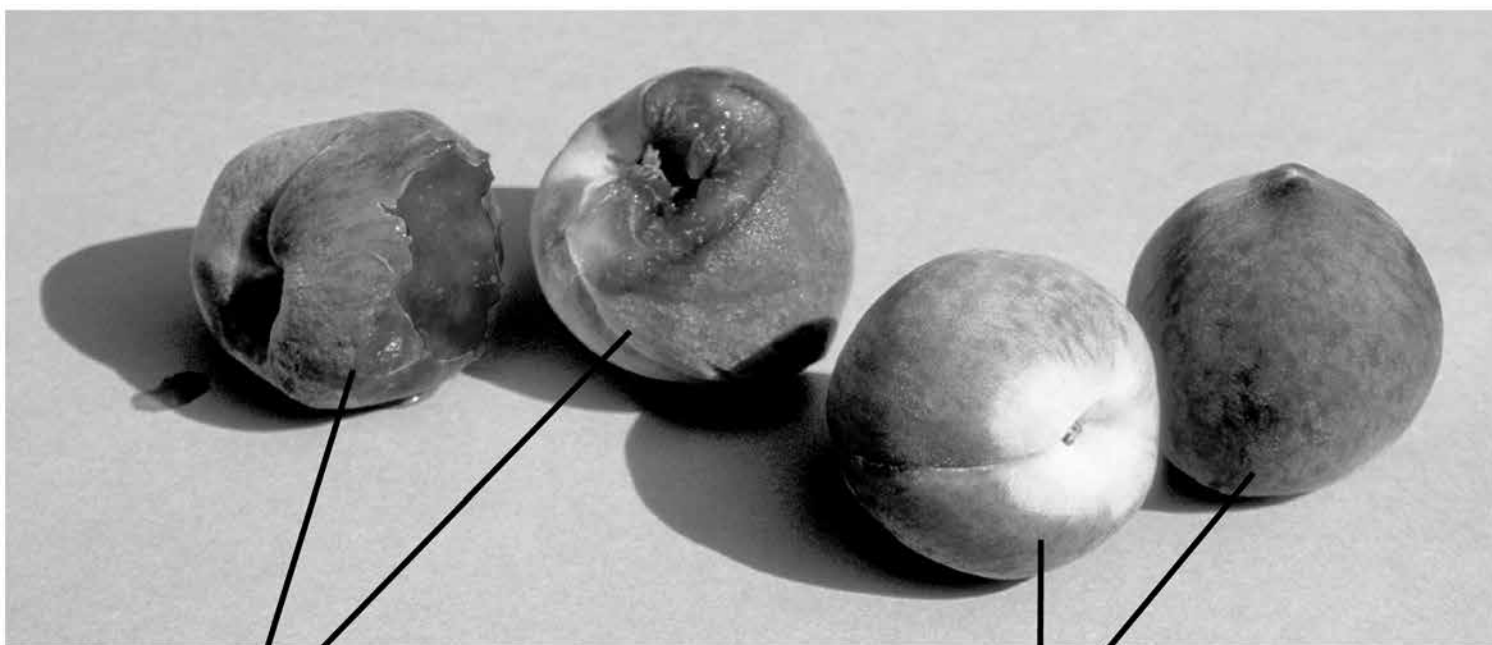
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Food can be irradiated to make it safer to eat.

FIGURE 4 shows a photograph of peaches.

Two of the peaches were irradiated.

The photograph was taken one week after irradiation.

FIGURE 4

**One week old,
NOT irradiated**

**One week old,
IRRADIATED**



0 5 . 3 Food is packaged and then irradiated.

**Explain why food is irradiated using gamma radiation rather than alpha or beta radiation.
[2 marks]**



37

0 5 . 4 Some people are concerned that irradiated food could be radioactive.

Describe how irradiated food is different from food that is radioactive. [2 marks]

9

[Turn over]



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0 6

Water travels through plants in xylem tissue.

0 6 . 1

Describe the structure of xylem tissue. [2 marks]

[Turn over]



40

FIGURE 5 shows guard cells around open stomata magnified 800 times.

FIGURE 5



41

0 6 . 2 Take the image size of one of the guard cells to be 26 millimetres long.

Calculate the real length of the guard cell in micrometres.

**Include the equation you are using to calculate your answer.
[3 marks]**

Real length of guard cell =
_____ micrometres

[Turn over]



0 6 . 3 Guard cells increase in volume and become curved to open stomata.

Explain how guard cells increase in volume. [2 marks]



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[Turn over]



06.4 The Baobab tree grows in Botswana, Africa.

The tree has no leaves for up to 9 months of the year.

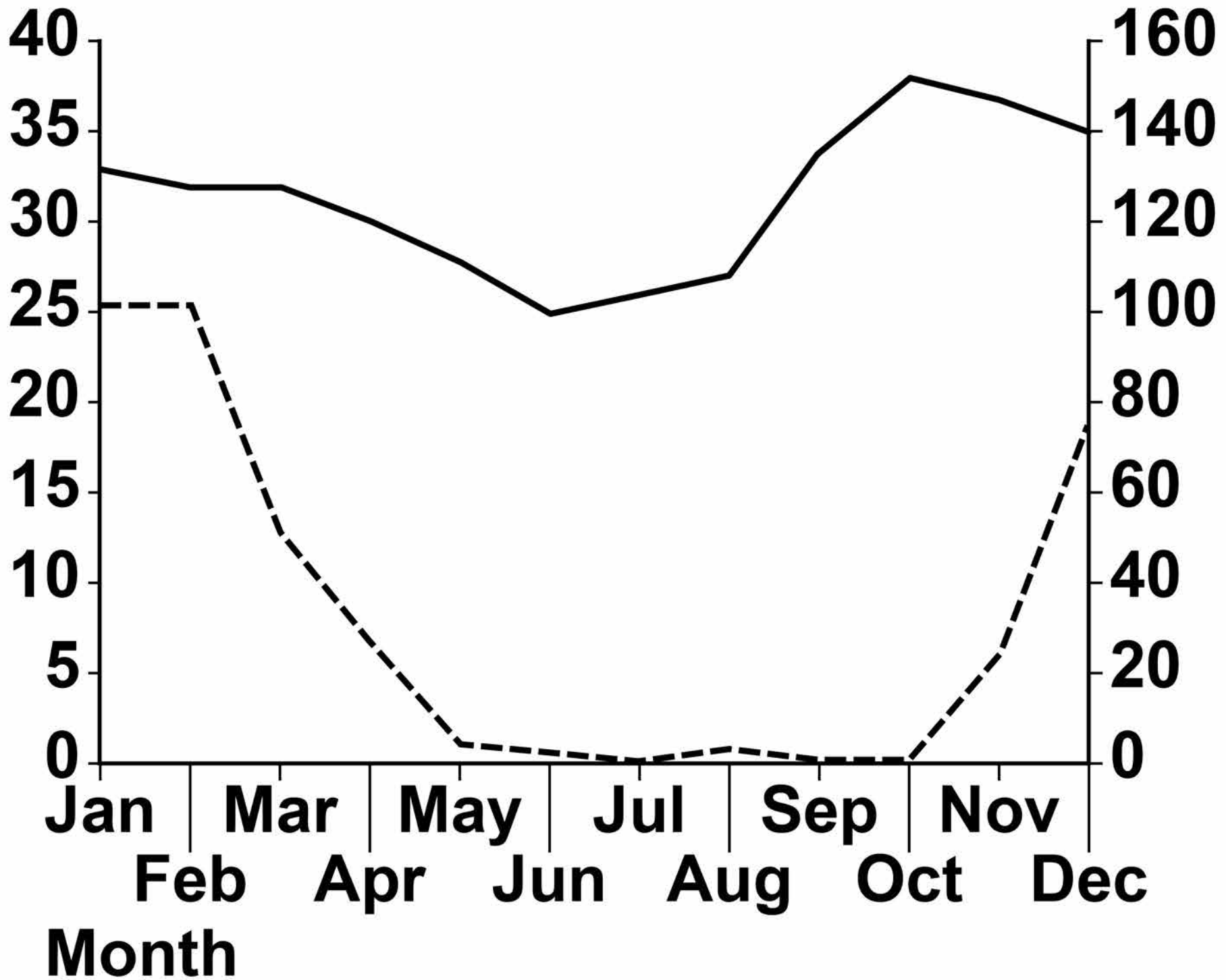
FIGURE 6, on page 45, shows the average temperature and rainfall each month in Botswana.



FIGURE 6

Temperature
in °C

Rainfall
in mm

**KEY**

———— Average maximum
temperature in °C

----- Average rainfall in mm

[Turn over]



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0	7
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Marfan syndrome is a rare genetic disorder that causes problems with many body systems.

0	7	.	1
---	---	---	---

Which sentence best describes a gene? [1 mark]

Tick ONE box.

A long chain of carbohydrate

A short section of DNA

All of the chromosomes in an organism

Several amino acids joined together



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07.2 What does a gene code for?
[1 mark]

Tick ONE box.

A carbohydrate polymer

A DNA double helix

One glycerol and three fatty acids

A sequence of amino acids

07.3 What scientific term is used to describe all the genes of one organism? [1 mark]

[Turn over]



50

07.4 What term is used to describe the observed characteristics of an individual? [1 mark]

Tick ONE box.

Allele

Genotype

Homozygous

Phenotype



51

07.5 Marfan syndrome is caused by a dominant allele, R.

The normal allele is recessive, r.

A man who is heterozygous for Marfan syndrome has a child with a woman who does not have the disorder.

Draw a genetic diagram to show the probability of their child inheriting Marfan syndrome.
[4 marks]

Probability = _____

[Turn over]



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[Turn over]



0	8
---	---

Some students investigated the compounds in a green lettuce leaf and a red cabbage leaf.

The students placed each leaf in boiling ethanol and then tested each leaf for starch.

0	8	.	1
---	---	---	---

The boiling point of ethanol is 78 °C

Ethanol is flammable so should not be directly heated with a Bunsen burner.

Give ONE way ethanol can be boiled safely.

Do NOT refer to wearing goggles in your answer. [1 mark]



0 8 . 2 Describe how the students could test the leaves for starch.

Give the result if starch is present. [2 marks]

Test _____

Result _____

[Turn over]



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[Turn over]



TABLE 5 shows the students' results. The distance the solvent and each pigment moved was measured from the start line.

TABLE 5

	Green lettuce		Red cabbage	
	Distance moved in mm	R _f value	Distance moved in mm	R _f value
Solvent front	120	–	113	–
Yellow-green pigment	18	0.15	14	0.12
Bright green pigment	24	0.20	Not found	Not found
Yellow pigment	40	0.33	46	0.41
Orange pigment	120	1.00	113	1.00



TABLE 6 shows the known R_f value ranges of some pigments.

TABLE 6

Pigment	R_f value range
Carotene	0.89 – 0.98
Pheophytin a	0.42 – 0.49
Pheophytin b	0.33 – 0.40
Chlorophyll a	0.24 – 0.30
Chlorophyll b	0.20 – 0.26
Xanthophyll	0.04 – 0.28

[Turn over]



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0 8 . 4 One pigment was found in the green lettuce leaf, but was NOT found in the red cabbage leaf.

Describe why it is NOT possible to be certain what this pigment is.

Use the information in TABLE 5 on page 58 and TABLE 6 on page 59 to help you. [1 mark]

[Turn over]



Repeat of TABLE 6

Pigment	R_f value range
Carotene	0.89 – 0.98
Pheophytin a	0.42 – 0.49
Pheophytin b	0.33 – 0.40
Chlorophyll a	0.24 – 0.30
Chlorophyll b	0.20 – 0.26
Xanthophyll	0.04 – 0.28

0 8 . 5 The experiment was repeated and the solvent front travelled 140 mm from the start line.

Calculate the range of distances where the pigment carotene would be seen.

Use the equation for calculating R_f values and the information in TABLE 6 to help you. [5 marks]



66

09

An understanding of relative size is essential in science.



67

0 9 . 1 Draw **ONE** line from each structure to the approximate radius of that structure.
[4 marks]

Structure**Approximate radius**

a bacterial cell

 $1 \times 10^{-14} \text{ m}$

a large molecule

 $5 \times 10^{-10} \text{ m}$

an animal cell

 $1 \times 10^{-10} \text{ m}$

an atom

 $1 \times 10^{-6} \text{ m}$ $2 \times 10^{-5} \text{ m}$ $3 \times 10^{-9} \text{ m}$ **[Turn over]**

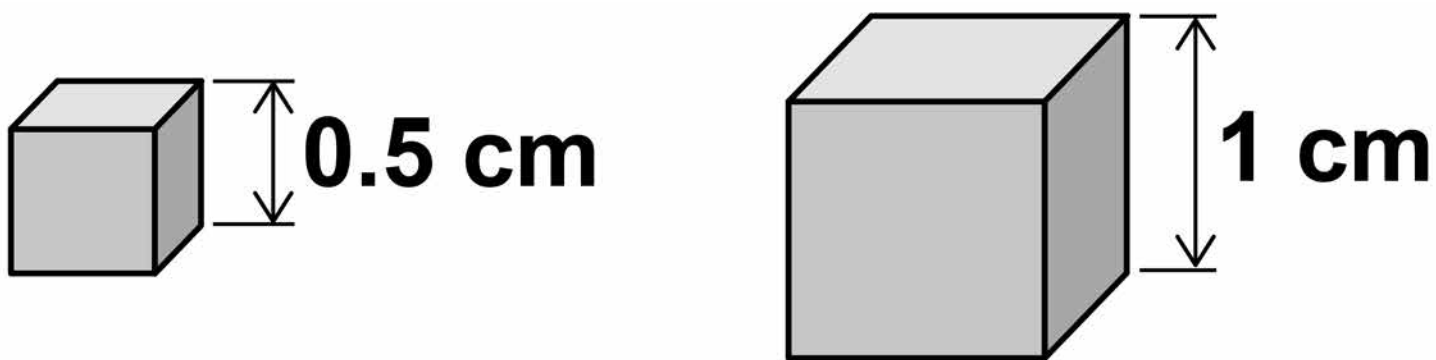
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FIGURE 7 shows two model cells.

Both models are cubes.

They are not drawn to scale.

FIGURE 7



0 9 . 3 Explain why a bacterium can rely on diffusion for gas exchange, but animals need a transport system. [3 marks]

There are no questions printed on this page

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Question	Mark
1	
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9	
TOTAL	

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