

Please write clearly in block capitals.	
Centre number	Candidate number
Surname	
Forename(s)	
Candidate signature	

Level 3 Certificate and Extended Certificate in Applied Science KEY CONCEPTS IN SCIENCE

Unit Number: J/507/6497

Monday 23 January 2017

Morning

Time allowed: 1 hour 30 minutes

Materials

For this paper you must have:

• a calculator.

Instructions

- Use black ink or black ball-point pen.
- Answer all questions.
- You must answer the questions in the spaces provided. Do not write outside the box around each page or on blank pages.
- Do all rough work in this book. Cross through any work you do not want to be marked.

Information

- You will be provided with a copy of the Periodic Table and formula sheet.
- There are three sections in this paper:

Section A – Biology

Section B - Chemistry

Section C - Physics.

- You should answer all questions in each section.
- You should spend approximately 30 minutes on each section.
- The marks for questions are shown in brackets.
- The maximum mark for this paper is 60.

Advice

· Read each question carefully.



Section A – Biology

			Answer all questions in this section	
Ir	n the m	nultip	ole choice questions, only one answer per question is allowed.	
F	or eac	h ar	nswer completely fill in the circle alongside the appropriate answer.	
С	ORREC	CT MI	WRONG METHODS ()	
If	you w	ant	to change your answer you must cross out your original answer as showr	n.
	you welect a		to return to an answer previously crossed out, ring the answer you now wown.	ish to
0	1	-	orts scientists study breathing and respiration. Breathing and respiration ferent processes.	are
		(a)	Select the correct feature of breathing.	
		Bre	eathing is:	
				[1 mark]
		A	physical and external.	
		В	chemical and external.	0
		С	physical and internal.	0
		D	chemical and internal.	0



	Respiration occurs in cells. Respiration has different stages. the stages of respiration.	Glycolysis is one of
	Where in a cell does glycolysis occur?	[1 mark]
Α	Cytoplasm	
В	Lysosomes	
С	Mitochondria	0
D	Nucleus	0
(c)	The Krebs cycle occurs in mitochondria. Figure 1 shows a m	nitochondrion.
	Figure 1	
	A	
the	ure 1 shows a mitochondrion that has been magnified 5000 till distance between point A and point B is 4.6 cm	-
the		
the	distance between point A and point B is 4.6 cm	efore magnification. [2 marks]
the	distance between point A and point B is 4.6 cm culate the length of the mitochondrion, in micrometres (µm), be	efore magnification. [2 marks]
the	distance between point A and point B is 4.6 cm culate the length of the mitochondrion, in micrometres (µm), be	efore magnification. [2 marks]
Cal	distance between point A and point B is 4.6 cm culate the length of the mitochondrion, in micrometres (µm), be	efore magnification. [2 marks]
Cal	distance between point A and point B is 4.6 cm culate the length of the mitochondrion, in micrometres (µm), be compared to the length of the mitochondrion in micrometres (µm), be compared to the length of the mitochondrion in micrometres (µm), be compared to the length of the mitochondrion in micrometres (µm), be compared to the length of the mitochondrion in micrometres (µm), be compared to the length of the mitochondrion in micrometres (µm), be compared to the length of the mitochondrion in micrometres (µm), be compared to the length of the mitochondrion in micrometres (µm), be compared to the length of the mitochondrion in micrometres (µm), be compared to the length of the	efore magnification. [2 marks]



(e)	A sports scientist is measuring the basal metabolic rate (BMR) of different people
	BMR is the minimum amount of energy required to maintain essential body
	processes. The sports scientist uses direct calorimetry to determine the BMR of the different people.

Describe how BMR is measured using direct calorimetry.	[4 marks]

(f) Another sports scientist measured the BMR of men and women using indirect calorimetry. The men and women in the study had different body masses. Table 1 shows the mean BMR values for different body masses.

Table 1

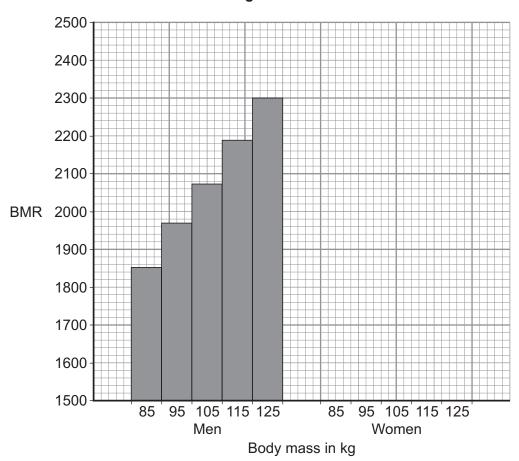
	В	/IR
Body mass in kg	Men	Women
85	1855	1550
95	1970	1620
105	2075	1700
115	2187	1775
125	2300	1855



Complete the bar chart in **Figure 2** to show the data for the women. The results for the men have been done for you.

[2 marks]





(g) Table 1 shows that BMR for men is higher than BMR for women of the same body mass.

Give one reason for this pattern.

[1 mark]

Turn over for the next question



for ped effi	e global human population is rapidly increasing. The production of sufficient food the increasing number of people is a major global concern. Food poverty is when ople do not have enough food for a healthy diet. Scientists try to increase the ciency of food chains and ecosystems to help reduce food poverty. The efficiency ferent ecosystems can be assessed by comparing primary production.
(a)	What is meant by gross primary production (GPP)? [1 mar
(b)	The GPP of a tropical rainforest is estimated to be 160 000kJ $\rm m^{-2}$ per year. The net primary production (NPP) of the tropical rainforest is 40 000 kJm ⁻² per year.
	Explain why the NPP is lower than the GPP. [2 mark]
_	
(c)	The NPP in a temperate rainforest in the UK is estimated to be 26 000 kJm ⁻² per year.
	Suggest three reasons why the NPP in a temperate forest is lower than in a tropical rainforest.
Re	ason 1
Re	ason 2
Re	ason 3



7

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(d) Some dieticians promote a meat-free diet. This would improve the efficiency of food chains.	
Give two reasons why the efficiency of a food chain would increase if more people ate a meat-free diet.	
[2 marks]	
Reason 1	
Reason 2	

Turn over for Section B



Section B – Chemistry

	Answer all questions in this section	
0 3	Material scientists consider the properties of a substance carefully before choosing it a particular function. The properties of a substance depend upon the type of bondin and the forces between the particles. A material scientist requires a substance to conduct electricity at 900°C, but be	
	an insulator in temperatures below 900°C. The material scientist selects an ionic substance.	
	(a) Draw a labelled diagram to show the ionic structure of solid sodium chloride. Your diagram should be a 3-dimensional representation. [2 mail]	rks]
	(b) Explain why sodium chloride will only conduct electricity at high temperatures. [2 mail]	rkel



	Exp	lain why magnesium oxide has a higher melting point than sodium c	hloride. [2 mar
			[2 IIIai
(d)	Met	al hydroxides are ionic substances.	
` ,	(i)	Give the formula for calcium hydroxide.	
			[1 m
	(ii)	Barium hydroxide will react in a similar way to calcium hydroxide.	
	(")	State why.	
		otate wily.	[1 m
	(iii)	Write the balanced symbol equation for the neutralisation of barium by hydrochloric acid (HCI).	hydroxi [2 ma



(a)	Sta	te in which block of the Periodic Table copper is located.
_		[1
(b)	-	oper is extracted from copper ore. High-grade copper ore consists mostly per carbonate (CuCO ₃).
	(i)	10 tonnes of copper ore contains 54% CuCO ₃ .
		Calculate the mass of copper carbonate in the ore. Give the correct unit your answer.
		[2 n
_		
_		
		Mace -
		Mass =
	(ii)	
	(ii)	$\label{eq:mass} \text{Mass} = \underline{\hspace{2cm}}$ During processing, the copper carbonate is reacted with sulfuric acid to no copper sulfate. $\text{CuCO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{CuSO}_4 + \text{H}_2\text{O} + \text{CO}_2$
	(ii)	During processing, the copper carbonate is reacted with sulfuric acid to n copper sulfate.
	(ii)	During processing, the copper carbonate is reacted with sulfuric acid to n copper sulfate. $ \text{CuCO}_3 \ + \ \text{H}_2\text{SO}_4 \ \rightarrow \ \text{CuSO}_4 \ + \ \text{H}_2\text{O} \ + \ \text{CO}_2 $
_	(ii)	During processing, the copper carbonate is reacted with sulfuric acid to recopper sulfate. $ \text{CuCO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{CuSO}_4 + \text{H}_2\text{O} + \text{CO}_2 $ The relative molecular mass (M _r) of copper carbonate is 123.5. Calculate the relative molecular mass (M _r) of copper sulfate (CuSO ₄).
	(ii)	During processing, the copper carbonate is reacted with sulfuric acid to recopper sulfate. $ \text{CuCO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{CuSO}_4 + \text{H}_2\text{O} + \text{CO}_2 $ The relative molecular mass (M _r) of copper carbonate is 123.5. Calculate the relative molecular mass (M _r) of copper sulfate (CuSO ₄).

		sulfate. Assume 100% yield.	[3 marks
			[O mark
_			
		Mass =	
(c)	Aqu	leous copper ions (Cu ²⁺) are a blue colour.	
	(i)	Give the electron arrangement of a Cu ²⁺ ion.	
	(-)		[1 mark
	(ii)	Explain why aqueous copper ions are blue. Use the idea of electron	
		transitions in your explanation.	[O]
			[2 marks



Section C - Physics

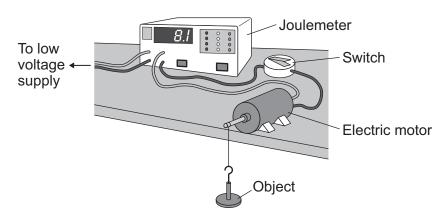
Answer all questions in this section

0 5

A technician thinks the efficiency of an electric motor depends on the mass being lifted. She wants to design an experiment to calculate the efficiency of an electric motor as it lifts different masses from the floor onto a laboratory bench. She has a joulemeter to measure the energy transferred by the motor as the mass is lifted.

Figure 3 shows some of the equipment she plans to use.





- (a) Describe how she would do this experiment. In your description you should state:
 - the measurements she would take
 - how she should ensure that her results are valid
 - how she should present her results.

[5 marks]



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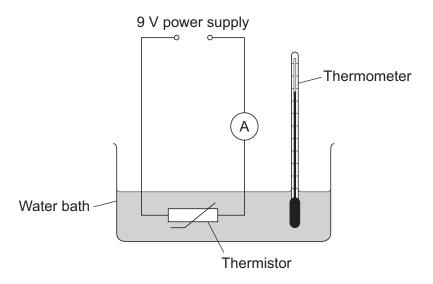
(b) (i)	A 400 g mass is lifted 0.85 m from the floor to the bench.
	Calculate the change in gravitational potential energy of the mass a Assume $g = 9.8 \text{ ms}^{-2}$.
	Change in gravitational potential energy =
(ii)	The joulemeter reading is 8.1 J
	Calculate the efficiency of the electric motor.
	Efficiency =
(c) Giv	re one reason why the efficiency of the electric motor is not 100%.



An apprentice engineer investigates the properties of a thermistor as its temperature changes. He uses a water bath to change the temperature of the thermistor.

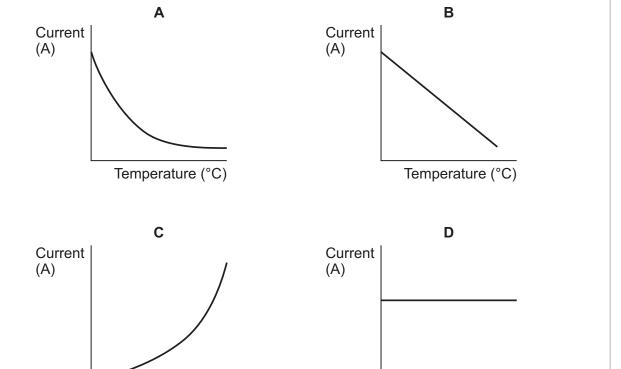
Figure 4 shows the apparatus used.

Figure 4



(a) (i) Which graph shows the change in current as the thermistor is heated?Circle the correct letter.

[1 mark]



Temperature (°C)

Turn over ▶

Temperature (°C)



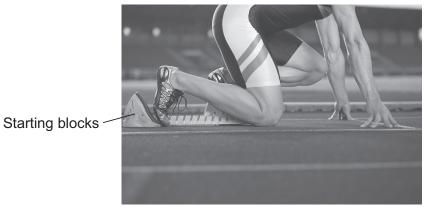
	[2 m
(b)	The apprentice engineer uses the thermistor to make a potential divider circuit.
(~)	Figure 5 shows the potential divider circuit.
	Figure 5
	9 V power supply
	A
	Fixed resistor
	The voltage across the thermistor decreases as the thermistor is heated.
	State what happens to the voltage across the fixed resistor.
	Give one reason for your answer. [2 m
(c)	Give one use for a potential divider circuit which includes a thermistor.
(0)	



o 7 Figure 6 shows a sprinter at the start of a race.

Figure 6

17



(a) Explain why the sprinter accelerates forwards when using the starting blocks.

Use Newton's Laws of Motion in your explanation.	[3 marks]



(b) At the start of the race the sprinter accelerates uniformly at 1.3 m s ⁻² .
	He reaches a maximum speed of 10.4 m s ⁻¹ .
	Calculate the distance he has run when he has reached his maximum speed. [2 marks]
-	
-	
-	
-	
	Distance = m

20

END OF QUESTIONS



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