

## **Cambridge Assessment International Education**

Cambridge Ordinary Level

CANDIDATE NAME					
CENTRE NUMBER			CANDIDATE NUMBER		

BIOLOGY 5090/32

Paper 3 Practical Test

May/June 2019

1 hour 15 minutes

Candidates answer on the Question Paper.

Additional Materials: As specified in the Confidential Instructions.

## **READ THESE INSTRUCTIONS FIRST**

Write your centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use an HB pencil for any diagrams or graphs.

Do not use staples, paper clips, glue or correction fluid.

DO NOT WRITE IN ANY BARCODES.

Answer all questions.

Write your answers in the spaces provided on the Question Paper.

Electronic calculators may be used.

You may lose marks if you do not show your working or if you do not use appropriate units.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [ ] at the end of each question or part question.

For Examiner's Use			
1			
2			
Total			

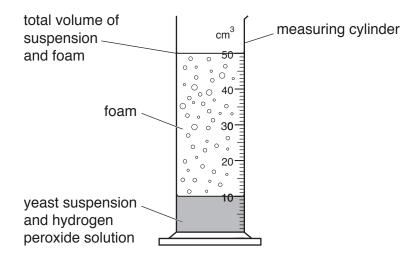
This document consists of **7** printed pages and **1** blank page.



## In order to plan the best use of your time, read through all the questions on this paper carefully before starting work.

1 Catalase is an enzyme found in living cells. This enzyme catalyses the breakdown of hydrogen peroxide into oxygen and water. A suspension of yeast cells in water can be used as a source of catalase.

The oxygen produced is given off as gas and causes foam to form as shown in the diagram. The larger the volume of foam produced, the greater the activity of the catalase.



You are going to investigate the activity of the enzyme catalase.

You are provided with a suspension of yeast cells in water as your source of catalase, and some hydrogen peroxide solution.

Use the following method:

- use a stirring rod to stir the yeast suspension in the beaker
- use a syringe to add 5 cm<sup>3</sup> of yeast suspension to the measuring cylinder
- use a clean syringe to add 5 cm<sup>3</sup> of hydrogen peroxide solution to the measuring cylinder
- immediately start timing
- after 1 minute measure the total volume of the suspension, including any foam formed on top, and record this in the table
- repeat the measurement after every minute for 5 minutes, recording the results in the table.

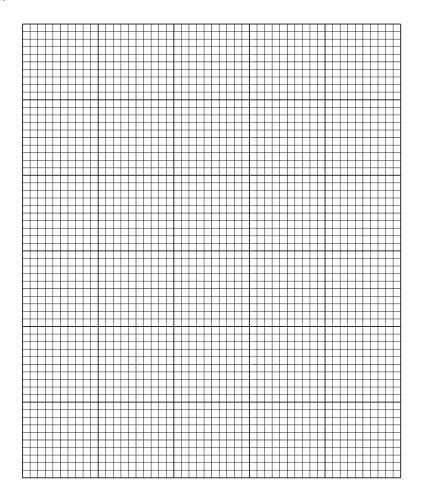
(a)

time / minutes	total volume of suspension and foam / cm <sup>3</sup>
0	10
1	
2	
3	
4	
5	

[2]

[4]

(b) (i) On the grid provided, plot a line graph of the data in the table. Join the points with ruled lines.



(ii)	Use the results to state the time period during which catalase activity was the greatest.
	[2]
(iii)	Describe the shape of the graph.

(c)	(i)	) Suggest reasons for								
		•	stirring the yeast suspension at the start of the experiment							
		•	using different syringes for measuring the yeast suspension and the hydrogen peroxide solution.							
			[2]							
	(ii)	Sug	gest a source of inaccuracy in the method for this experiment.							
			[1]							
(d)			n experiment based on the method already described (see page 2) to investigate the							
	effe	ct of	increasing temperature on the activity of catalase. Give full experimental details.							
			[6]							
			[Total: 19]							

© UCLES 2019 5090/32/M/J/19

Question 2 starts on page 6.

2 The photograph shows an apple cut in half.



(a) In the space below make a large drawing of the apple as it appears in the photograph. Label a seed on your drawing.

[5]

(b) (i) Draw a straight line between the lines labelled **A** and **B** on the photograph. Measure and record the length of your line.

measurement ......mm [1]

(ii) Draw a straight line in a similar position on your drawing. Measure and record the length of your line.

measurement ...... mm [2]

(	iii)	Use your measure the apple as it ap	drawing comp	ared with				
	$\mbox{magnification} \times \label{eq:magnification}$ A student carried out three tests on the apple, one with iodine solution, one with biuret solution							
	(i)	d one with Benedict's solution, to discover its nutrient content.  State which test required the use of heat.  [1]						
	(ii) The apple tested positive with Benedict's solution and negative with both biuret solution and iodine solution. Use this information to complete the table with the studen observations at the end of the tests.							
		test solution	observation	conclusion				
	В	enedict's solution		positive				
		biuret solution		negative				
		iodine solution		negative	[3]			
(iii) State what these results tell you about the nutrient content of the apple.								
(d)	(d) (i) Describe how the student could test the apple for the presence of fat. Include the observation for a positive result.							
			[3]					
(ii) State and explain a safety precaution the student must take when carrying out test in (d)(i).								
	precaution							
explanation								
					[2]			

[Total: 21]

8

## **BLANK PAGE**

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (UCLES) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

To avoid the issue of disclosure of answer-related information to candidates, all copyright acknowledgements are reproduced online in the Cambridge Assessment International Education Copyright Acknowledgements Booklet. This is produced for each series of examinations and is freely available to download at www.cambridgeinternational.org after the live examination series.

Cambridge Assessment International Education is part of the Cambridge Assessment Group. Cambridge Assessment is the brand name of the University of Cambridge Local Examinations Syndicate (UCLES), which itself is a department of the University of Cambridge.

© UCLES 2019 5090/32/M/J/19