# CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

# MARK SCHEME for the May/June 2013 series

## **5054 PHYSICS**

**5054/22** Paper 2 (Theory), maximum raw mark 75

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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Page 2	Mark Scheme	Syllabus	Paper
	GCE O LEVEL – May/June 2013	5054	22

### Section A

1	(a)	trav	vels further in each second / in same time / between images	B1	
	(b)		) <i>d/t</i> in any form algebraic or numerical cm/s; 0.4(0) m/s	C1 A1	
	(c)		resistance increases ght constant	B1 B1	
	(d)	or I	ces balance /cancel no resultant/net force resultant of any two forces equal and opposite to third	B1	[6]
2	(a)		ce × distance pendicular distance	M1 A1	
	(b)	(i)	$T \times 8$ or $2000 \times 2$ seen 500 N	C1 A1	
		(ii)	(two forces) equal (in magnitude) (two forces) opposite (in direction)	B1 B1	[6]
3	(a)	(i)	( <i>W</i> =) <i>Fd</i> or $90 \times 0.3$ or $90 \times 30$	C1	
			27 J	A1	
		(ii)	(P=) W/t or Fd/t or 27(× 20)/60 or 27/3	C1	
			9(.0) W	A1	
	(b)	(i)	800 × 30/180 or 800/6 or 6 seen or proportionality clearly used	C1	
			133 <b>or</b> 130 cm	A1	
		(ii)	extension more than 143 cm <b>or</b> (extra) extension > 10 cm <b>or</b> (some) extension permanent	B1	[7]

Page 3	Mark Scheme	Syllabus	Paper
	GCE O LEVEL – May/June 2013	5054	22

**B**1 4 (a) gamma rays, visible light, infra-red (b) (i) microwaves **B1** (ii) satellite (receives and) sends/transmits/emits/boosts/amplifies signal **B1** (iii) cover a large area over the horizon / only one (transmitter/station) needed etc. B1 [4] or unaffected by tall buildings/hills or no obstructions 5 (a) electrons move onto polythene / rod **B**1 **B1** electrons/negative charge move off cloth (b) (region of space) where force is exerted on a charge **B1** (c) (i) unlike charges attract **B**1 or (rod) attracts +ve charge/ions/particles repels like charge **B**1 or (rod) repels –ve charge/ions/electrons/particles (ii) (net) positive charge on water near rod **B**1 [6] **B1** 6 (a) (i) 2 to 2.1 (V) to any value between 11 and 12 (V) **or** above 2/2.1(V) (ii) temperature increases / gets hotter **B**1 (b) (i) (rate of) flow of charge/electrons **B1** (ii) 0.35 A cao **B**1 (iii) (I=) V/R algebraic C1 or 6/20 or 6/0.35 C1 0.3(0)(A)or  $1/R_T = 1/20 + 1/17.1$ **or** ( $R_T$  =) 9.2 (Ω) seen 0.65 A **A1** [7]

	Page	4	Mark Scheme	Syllabus	Paper	
			GCE O LEVEL – May/June 2013	5054	22	
7			le through or near A centered on or near X e arrow on line(s) around X <b>and</b> none wrong		B1 B1	
	<b>(b)</b> fie	ld <b>s</b> (dı	ue to X and Y) cancel <b>or</b> X and Y fields equal and o	pposite	B1	
	(c) (i)		ne left owards X/A/B		B1	
	(ii)		ent (in wire Y) and (magnetic) <u>field</u> (caused by otherwo (magnetic) fields interact	r wire)	B1	[5]
8	EITHE	R				
	(a) ste	eel / m	agnadur / alnico / magnetite		B1	
	(b) (i)	men	tion of cutting (lines of) magnetic field / change in (r	magnetic) flux	M1	
		<b>or</b> fa	at(est) rate of change ast(est) cutting ther explanation involving time		A1	
	(ii)	or tu	cal/upright urned through 90° ormal to (magnetic) field		B1	
	OR					
	(a) NO or	OT (ga invert	te) er		B1	
	<b>(b)</b> 1,0	)			B1	
	(c) (i)	(volt	age across R <sub>1</sub> ) <u>becomes</u> 0/low		B1	
	(ii)	decr	rease any of R <sub>1</sub> , R <sub>2</sub> , C <sub>1</sub> , C <sub>2</sub>		B1	[4]
				[Total:	45]	

Page 5	Mark Scheme	Syllabus	Paper
	GCE O LEVEL – May/June 2013	5054	22

### Section B

9	(a)	(air	) molecules hit walls / liquid (surface)	B1	
		(air	) molecules move fast(er) /great(er) kinetic energy	B1	
		•	) molecules hit <u>more</u> often/ <u>more</u> frequently/ <u>greater</u> rate / hard <u>er</u> / <u>more</u> force (liquid) molecules evaporate	B1	[3]
	(b)	(i)	(flask) <u>in</u> (pure) <u>melting</u> ice (and water)	B1	
			(flask) in (pure) boiling water / above boiling water (at one atmosphere)	B1	
		(ii)	thin(ner) tube  or large(r) flask  or more air/less liquid  or use liquid that expands more (1 mark for each)	B2	
			of use liquid that expands more (1 mark for each)	DZ	
		(iii)	divisions not equally spaced <b>or</b> scale not uniform/not proportional	C1	
			different distance (along scale) for same temperature rise or different change in temperature for same distance (along scale)	A1	[6]
	(c)	(i)	( <i>M</i> =) $d \times V$ in any form <b>or</b> $1200 \times 5 \times 10^{-5} \times 0.15$	C1	
			$9(.0) \times 10^{-3} \text{ kg}; 0.009(0) \text{ kg}$	A1	
		(ii)	0.09(0) N ecf (i)	B1	
		(iii)	(P=) hdg in any form or (P=) F/A in any form	C1	
			1800 Pa	A1	[5]
	(d)		ids expand less (than air) great(er) forces between liquid molecules	B1	[1]
				[Total:	15]
10	(a)	cor	rect normal by eye rect angle of incidence between candidate's normal and incident ray rect angle of refraction marked between candidate's normal and BC	B1 B1 B1	[3]
	(b)	dec	crease / change in speed / wavelength	B1	[1]

Page 6			Paper	
		GCE O LEVEL – May/June 2013 5054	22	
(c)	) n=	sin i/sin r seen in any form	C1	
	$(\sin r =) \sin 45^{\circ}/1.5$ <b>or</b> 0.47(14) seen			
	28(	.1)°	C1	[3]
(d)	(d) refracts less at first face and on correct side of normal			
		raction at second face away from normal so that red ray and blue ray are erging	B1	[2]
(e)	) (i)	angle of incidence is 0  or ray along normal/perpendicular to glass	B1	
	(ii)	angle of incidence/ $\theta$ is larger than critical angle total internal reflection occurs	B1 B1	
	(iii)	reflected ray drawn correctly <b>and</b> emerging without refraction from block	B1	
	(iv)	(eventually) light emerges (into air at Q)  or light refracts (out at Q)  or (weak) refracted ray appears	B1	
		light emerging at Q coloured in some way ${f or}$ correct description of movement of reflected ray (as $ heta$ decreases)	B1	[6]
			[Total:	15]
11 (a)		ver supply with ammeter and heater in series tmeter in parallel with heater/ power supply	B1 B1	[2]
<b>(b</b> )	) (i)	$(P=)VI$ in any form or $4.2 \times 12$	C1	
		50(.4) W	A1	
	(ii)	( <i>E</i> =) $Pt$ i.e. any power × any time e.g. $50(.4) \times 8$	C1	
		8/60 or 0.13(3) seen or division by 1000 seen anywhere	C1	
		0.0067(2) (kWh)	A1	[5]

Page 7	Mark Scheme	Syllabus	Paper
	GCE O LEVEL – May/June 2013	5054	22

(c) (i) molecules escape (from surface/leave water) / become gas or vapour / break bonds

C1

fast(er) moving / high energy/ energetic molecules escape

Α1

(ii)

change M1	explanation A1	
wind / draught / breeze	wind knocks molecules away	
or larger surface area	more chance/possibility of escape/more space to escape or more molecules come to/near/at surface	
or decrease humidity / drier air	fewer molecules return/from air	
or decrease atmospheric pressure	fewer air molecules to hit during escape	

(iii) evaporation occurs at surface and boiling inside liquid/bubbles evaporation occurs at any temperature (accept room temperature) and boiling occurs at boiling point/100°C/ fixed / specific temperature evaporation increased by draughts/higher temp/more area and boiling is not OR increase in pressure stops boiling but only reduces evaporation any two

B2 [6]

(d) water heats air (by conduction)

В1

or water loses heat/energy (to cup or air)
or air gains heat/energy (from water)

hot / heated air / particles rise

B1 [2]

or cold air / particles sink

or hot air is less dense

or cold air is more dense

[Total: 15]