



Cambridge International Examinations
Cambridge Ordinary Level

COMBINED SCIENCE

5129/12

Paper 1 Multiple Choice

October/November 2015

1 hour

Additional Materials: Multiple Choice Answer Sheet
Soft clean eraser
Soft pencil (type B or HB is recommended)



READ THESE INSTRUCTIONS FIRST

Write in soft pencil.

Do not use staples, paper clips, glue or correction fluid.

Write your name, Centre number and candidate number on the Answer Sheet in the spaces provided unless this has been done for you.

DO NOT WRITE IN ANY BARCODES.

There are **forty** questions on this paper. Answer **all** questions. For each question there are four possible answers **A, B, C** and **D**.

Choose the **one** you consider correct and record your choice in **soft pencil** on the separate Answer Sheet.

Read the instructions on the Answer Sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer.

Any rough working should be done in this booklet.

A copy of the Periodic Table is printed on page 20.

Electronic calculators may be used.

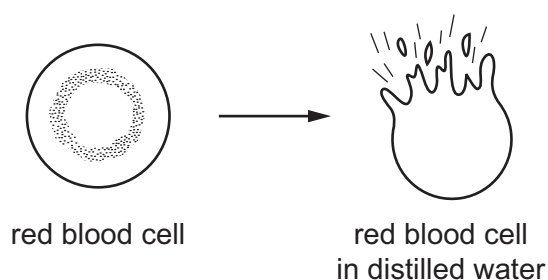
This document consists of **15** printed pages and **1** blank page.



- 1 Which pair of statements best explains why plant cells are more angular in shape than animal cells?

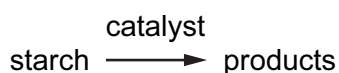
	plant cells	animal cells
A	cell wall present	cell wall absent
B	chloroplast present	chloroplast absent
C	one or more large vacuoles	small or no vacuoles
D	thin layer of cytoplasm	dense cytoplasm throughout cell

- 2 The diagram shows a red blood cell and the same cell after it has been placed in distilled water.



The change in appearance of the cell is because in distilled water

- A** the cytoplasm shrinks.
B water enters the cell by ingestion.
C water enters the cell by osmosis.
D water leaves the cell by diffusion.
- 3 The following reaction occurs in the human alimentary canal.



What is the catalyst and what is one of the products?

	catalyst	product
A	acid	glucose
B	acid	maltose
C	amylase	glucose
D	amylase	maltose

4 Which row describes the process of photosynthesis?

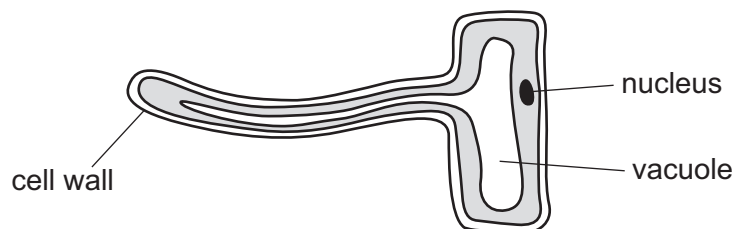
	raw materials	form of energy trapped by chlorophyll	products
A	carbon dioxide and water	nuclear	glucose and oxygen
B	carbon dioxide and water	light	glucose and oxygen
C	oxygen and water	nuclear	glucose and carbon dioxide
D	oxygen and water	light	glucose and carbon dioxide

5 A student changes his diet to one that combines very high fibre with very low fat.

Which row shows the likely outcome of his new diet?

	weight	constipation
A	decreases	decreases
B	decreases	increases
C	increases	decreases
D	increases	increases

6 The diagram shows a root hair cell.

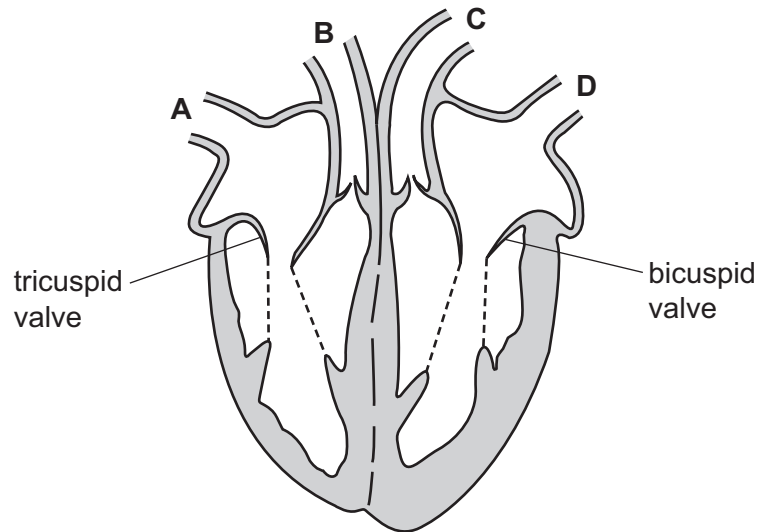


Which feature of a root hair cell does **not** affect the absorption of water?

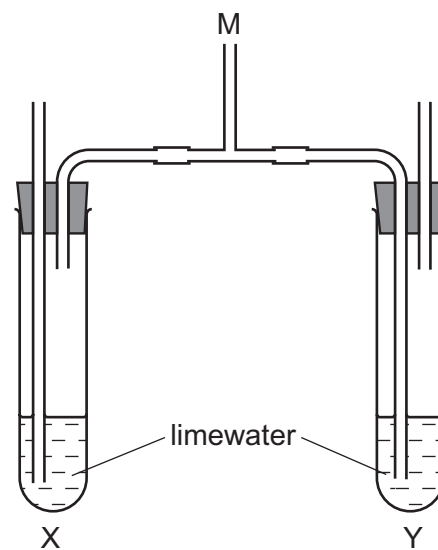
- A** The root hair cell has a cell wall.
- B** The root hair cell has a large surface area.
- C** The root hair cell has a nucleus.
- D** The root hair cell has a vacuole.

7 The diagram shows the heart.

Which vessel is an artery carrying deoxygenated blood?



8 The apparatus shown is used to investigate gas exchange during breathing.



Which describes the appearance of solutions X and Y after a person has exhaled several times through tube M?

	solution X	solution Y
A	clear	clear
B	clear	cloudy
C	cloudy	clear
D	cloudy	cloudy

- 9 Excretion is the removal of waste products of metabolism.

Which row matches the waste products water, urea and carbon dioxide to the organ that removes them from the body?

	water	urea	carbon dioxide
A	kidney	kidney	lungs
B	kidney	liver	lungs
C	lungs	kidney	liver
D	lungs	liver	liver

- 10 What is the target organ of a hormone?

- A** the gland that destroys the hormone
- B** the gland that secretes the hormone
- C** the organ that destroys the hormone
- D** the organ where the hormone acts

- 11 It is estimated that 13.5 million people use opium-like substances such as heroin. Of these, 9.2 million people use heroin.

What is the percentage of users of opium-like substances who use heroin?

- A** 33% **B** 50% **C** 68% **D** 75%

- 12 Which natural process **removes** carbon dioxide from the air?

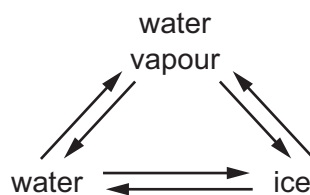
- A** decay
- B** digestion
- C** photosynthesis
- D** respiration

- 13 A young man has gone to the doctor with pain in passing urine and an unpleasant discharge from his penis. The doctor suspects he has contracted an infectious disease.

Which further information might the doctor seek to confirm the diagnosis?

- A** factual details of his recent sexual activity
- B** his blood sugar levels
- C** major recent changes to his diet or anything unusual he has recently eaten
- D** the young man's blood pressure

14 In which conversion do water molecules lose speed?



- A ice $\rightarrow$ water
- B ice $\rightarrow$ water vapour
- C water vapour $\rightarrow$ ice
- D water $\rightarrow$ water vapour

15 Which row shows an atom containing 24 neutrons?

	proton number	nucleon number
A	8	16
B	12	24
C	21	45
D	24	52

16 The table shows the electronic structure of four elements.

element	electronic structure
1	2,1
2	2,7
3	2,8,2
4	2,8,6

Which two elements form an ionic compound with the formula XY_2 ?

- A 1 and 2
- B 1 and 3
- C 2 and 3
- D 3 and 4

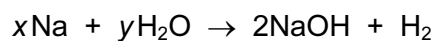
- 17 The table shows some properties of two solid compounds, X and Y.

property	X	Y
low melting point	yes	no
soluble in water	no	yes
conducts electricity when molten	no	yes

Which row describes the bonding in X and Y?

	X	Y
A	covalent	covalent
B	covalent	ionic
C	ionic	covalent
D	ionic	ionic

- 18 The equation shows the reaction between sodium and water.



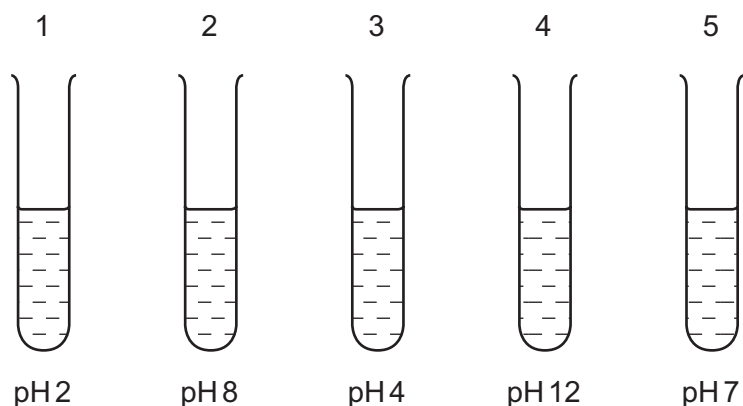
What are the values of x and y for the equation to be balanced?

	x	y
A	1	1
B	1	2
C	2	1
D	2	2

- 19** A student tests five solutions to discover which solutions were acidic, alkaline or neutral.

The student tests each solution using Universal Indicator paper to determine their pH.

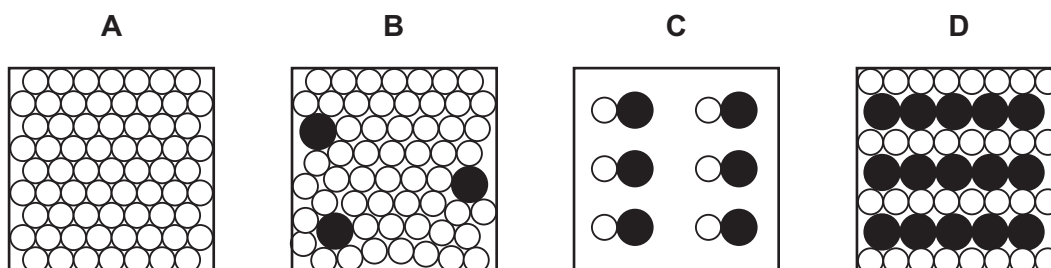
The results are shown below.



Which solutions are acidic?

- A** 1 and 3 **B** 1 and 5 **C** 2 and 3 **D** 2 and 4
- 20** Which statement about elements in a period of the Periodic Table is correct?
- A** Atoms of elements in the same period have the same number of electron shells.
- B** Atoms of elements in the same period have the same number of protons.
- C** Elements in the same period have similar chemical properties.
- D** Elements on the right of a period are more metallic than elements on the left of a period.
- 21** Brass is an alloy. Brass contains copper and zinc.

Which structure represents the arrangement of atoms in brass?



22 Q, R, S and T are four metals.

T reacts slowly with hydrochloric acid.

Q does not react with acid.

R reacts with steam but not with cold water.

S reacts violently with cold water.

What is the order of reactivity of the four metals, most reactive first?

A $Q \rightarrow T \rightarrow R \rightarrow S$

B $Q \rightarrow R \rightarrow T \rightarrow S$

C $S \rightarrow Q \rightarrow R \rightarrow T$

D $S \rightarrow R \rightarrow T \rightarrow Q$

23 Which of the three pollutants in air make rain water acidic?

1 carbon monoxide

2 nitrogen oxide

3 sulfur dioxide

A 1 and 3

B 2 and 3

C 2 only

D 3 only

24 Which pair of substances does **not** produce hydrogen gas?

A calcium and sulfuric acid

B copper and water

C sodium and water

D zinc and hydrochloric acid

25 Some properties of an organic compound, X, are listed.

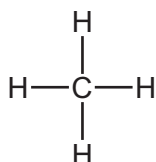
X is a saturated hydrocarbon.

X is a gas at room temperature.

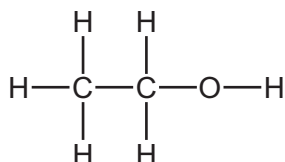
X can be cracked to produce ethene and hydrogen.

Which structure is X?

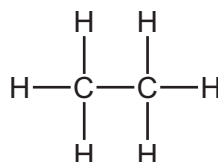
A



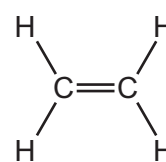
B



C



D



26 Bitumen is obtained from petroleum by fractional distillation.

What is a use of bitumen?

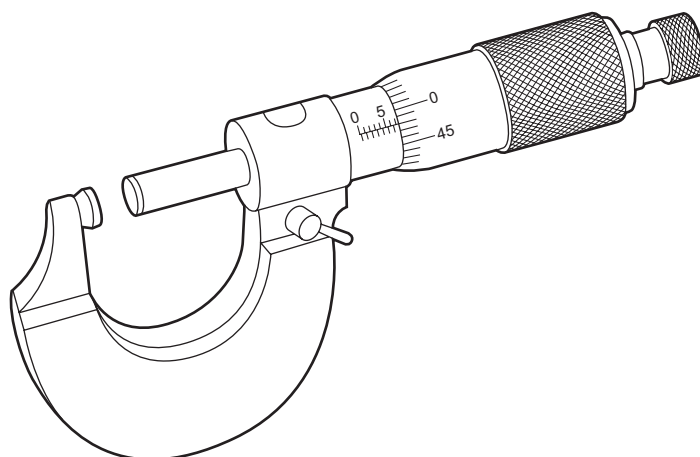
- A fuel for aircraft
- B fuel for cars
- C lubricants
- D making roads

27 Ethene is a member of the alkene homologous series.

Which statement about alkene molecules is correct?

- A They are able to add to each other to produce polymers.
- B They are saturated.
- C They contain carbon, hydrogen and oxygen atoms.
- D They only have single bonds between the atoms.

28 The diagram shows an instrument used in Physics.



What is the name of this instrument and what is it used to measure?

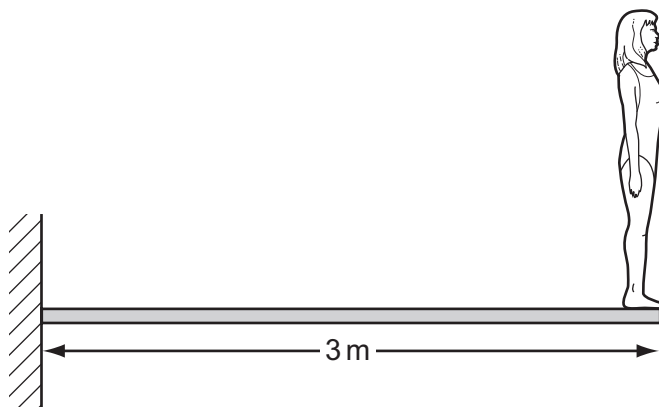
	name	used to measure
A	calipers	length
B	calipers	pressure
C	micrometer	length
D	micrometer	pressure

29 A force is applied to a moving object.

Which effect **cannot** occur?

- A Its direction changes.
- B Its velocity remains constant.
- C The object slows down.
- D The object speeds up.

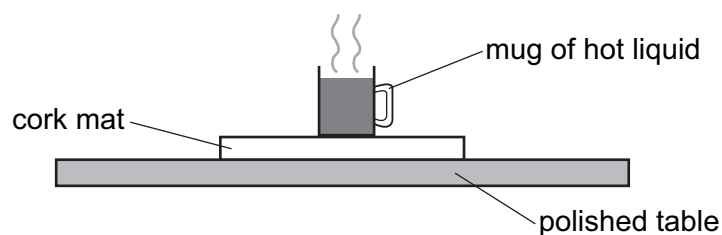
30 A diver, weighing 720 N, stands at the end of a springboard measuring 3 m long.



What is the moment about the support due to the weight of the diver?

- A 720 Nm B 720×3 Nm C $\frac{3}{720}$ Nm D $\frac{720}{3}$ Nm

31 To protect a polished table, a cork mat may be put on the table underneath a mug containing hot liquid.



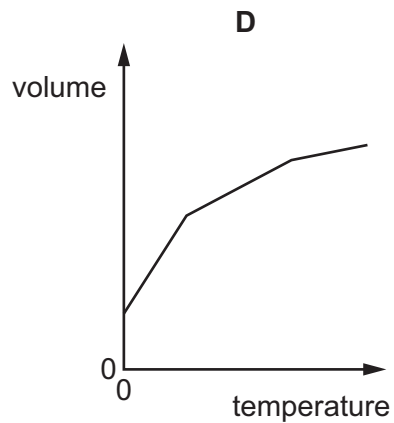
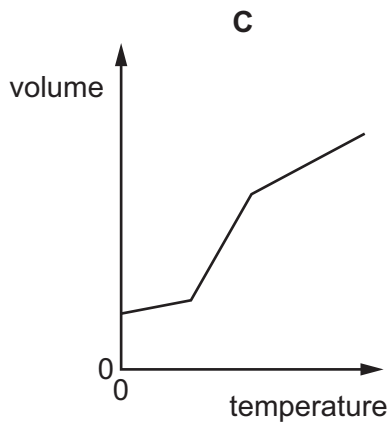
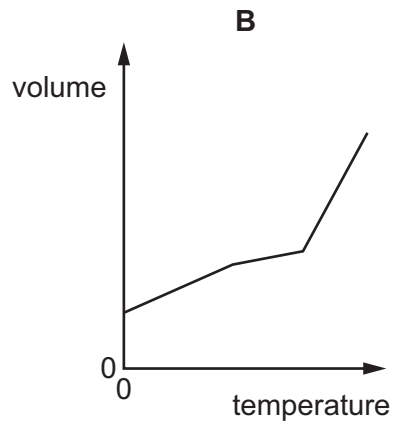
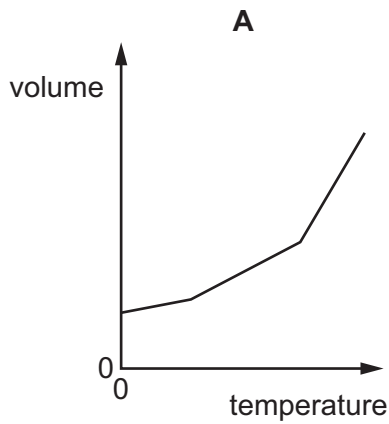
Why is this effective?

- A Cork is a good conductor.
- B Cork is a good radiator.
- C Cork is a poor conductor.
- D Cork is a poor radiator.

32 A piece of solid material is heated continually so that it melts and then boils.

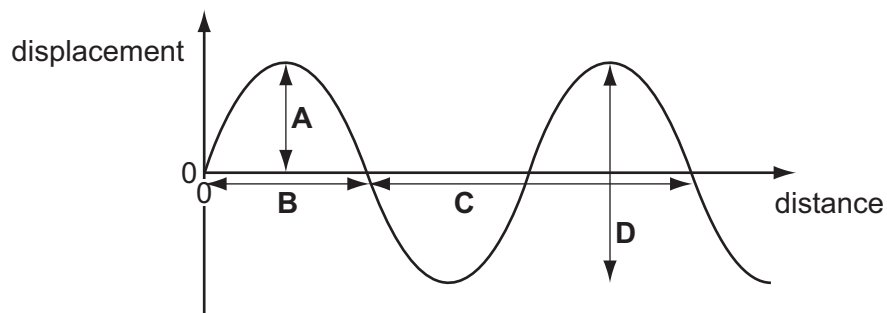
During the solid, liquid and gaseous states thermal expansion of the material occurs.

Which graph best shows the thermal expansion during each state?

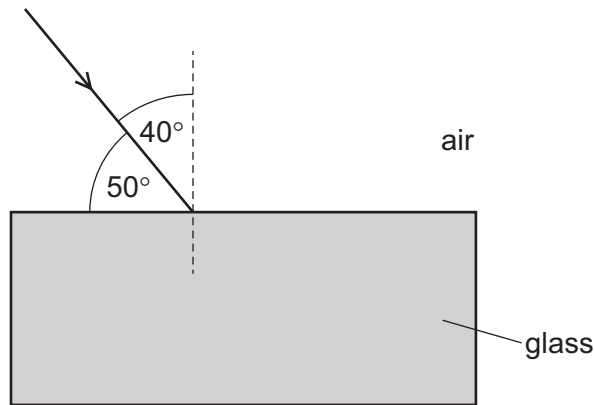


33 The diagram shows the displacement across a wave.

Which value is multiplied by the frequency to give the speed of the wave?



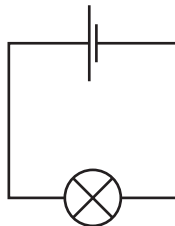
- 34 A ray of light is incident on the surface of a block of glass.



The refractive index of the glass is 1.5.

What is the angle of refraction of the ray in the glass?

- A** 25° **B** 27° **C** 31° **D** 33°
- 35 Which type of electromagnetic radiation travels at the highest speed through a vacuum?
- A** gamma-rays
B light waves
C radio waves
D none – all have the same speed
- 36 In the circuit shown, 20 J of energy is dissipated by the cell in driving 8.0 C of charge round the circuit.



What is the value of the e.m.f. of the cell?

- A** 0.4 V **B** 2.5 V **C** 28 V **D** 160 V

37 An electric kettle has a metal case.

An earth wire is connected to the metal case.

A fuse in the plug has a 13 A fuse rating.

A fault occurs and the live wire touches the metal case.

Which statement is **not** correct?

- A The current in the earth wire is greater than 13 A.
- B The current in the live wire is greater than 13 A.
- C The fuse keeps the current at 13 A.
- D The fuse begins to melt.

38 Which statement about the action of a transformer is correct?

- A An e.m.f. is induced in the secondary coil when an alternating voltage is applied to the primary coil.
- B An e.m.f. is induced in the secondary coil when there is a steady direct current in the primary coil.
- C The current in the secondary coil is always larger than the current in the primary coil.
- D The voltage output of the secondary coil is always larger than the voltage input of the primary coil.

39 Which particle is positively-charged?

- A electron
- B neutral atom
- C neutron
- D proton

40 Which row is correct for a beta-particle?

	the nature of a beta-particle	a beta-particle is stopped by
A	electromagnetic wave	a piece of paper
B	electromagnetic wave	twenty centimetres of aluminium
C	electron	a piece of paper
D	electron	twenty centimetres of aluminium

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DATA SHEET
The Periodic Table of the Elements

Group																										
I	II											III	IV	V	VI	VII	0									
												1 H Hydrogen 1														
7 Li Lithium 3	9 Be Beryllium 4											11 B Boron 5	12 C Carbon 6	14 N Nitrogen 7	16 O Oxygen 8	19 F Fluorine 9	20 Ne Neon 10									
23 Na Sodium 11	24 Mg Magnesium 12											27 Al Aluminium 13	28 Si Silicon 14	31 P Phosphorus 15	32 S Sulfur 16	35.5 Cl Chlorine 17	40 Ar Argon 18									
39 K Potassium 19	40 Ca Calcium 20	45 Sc Scandium 21	48 Ti Titanium 22	51 V Vanadium 23	52 Cr Chromium 24	55 Mn Manganese 25	56 Fe Iron 26	59 Co Cobalt 27	59 Ni Nickel 28	64 Cu Copper 29	65 Zn Zinc 30	70 Ga Gallium 31	73 Ge Germanium 32	75 As Arsenic 33	79 Se Selenium 34	80 Br Bromine 35	84 Kr Krypton 36									
85 Rb Rubidium 37	88 Sr Strontium 38	89 Y Yttrium 39	91 Zr Zirconium 40	93 Nb Niobium 41	96 Mo Molybdenum 42	101 Ru Ruthenium 44	101 Ru Ruthenium 44	103 Rh Rhodium 45	106 Pd Palladium 46	108 Ag Silver 47	112 Cd Cadmium 48	115 In Indium 49	119 Sn Tin 50	122 Sb Antimony 51	128 Te Tellurium 52	127 I Iodine 53	131 Xe Xenon 54									
133 Cs Caesium 55	137 Ba Barium 56	139 La Lanthanum 57	178 Hf Hafnium 72	181 Ta Tantalum 73	184 W Tungsten 74	186 Re Rhenium 75	190 Os Osmium 76	192 Ir Iridium 77	195 Pt Platinum 78	197 Au Gold 79	201 Hg Mercury 80	204 Tl Thallium 81	207 Pb Lead 82	209 Bi Bismuth 83	210 Po Polonium 84	210 At Astatine 85	210 Rn Radon 86									
87 Fr Francium	88 Ra Radium	89 Ac Actinium																								
58-71 Lanthanoid series																										
90-103 Actinoid series																										
													140 Ce Cerium 58	141 Pr Praseodymium 59	144 Nd Neodymium 60	150 Sm Samarium 62	152 Eu Europium 63	157 Gd Gadolinium 64	159 Tb Terbium 65	162 Dy Dysprosium 66	165 Ho Holmium 67	167 Er Erbium 68	169 Tm Thulium 69	173 Yb Ytterbium 70	175 Lu Lutetium 71	
													232 Th Thorium 90	232 Pa Protactinium 91	238 U Uranium 92	238 Np Neptunium 93	238 Pu Plutonium 94	238 Am Americium 95	238 Cm Curium 96	238 Bk Berkelium 97	238 Cf Californium 98	238 Es Einsteinium 99	238 Fm Fermium 100	238 Md Mendelevium 101	238 No Nobelium 102	238 Lr Lawrencium 103
Key													a X b a = relative atomic mass X = atomic symbol b = proton (atomic) number													

*58-71 Lanthanoid series
†90-103 Actinoid series

a	X	a = relative atomic mass
b	X	X = atomic symbol
		b = proton (atomic) number

The volume of one mole of any gas is 24 dm³ at room temperature and pressure (r.t.p.).