



Cambridge International Examinations
Cambridge Ordinary Level

COMBINED SCIENCE

5129/11

Paper 1 Multiple Choice

May/June 2017

1 hour

Additional Materials: Multiple Choice Answer Sheet
Soft clean eraser
Soft pencil (type B or HB is recommended)



READ THESE INSTRUCTIONS FIRST

Write in soft pencil.

Do not use staples, paper clips, glue or correction fluid.

Write your name, Centre number and candidate number on the Answer Sheet in the spaces provided unless this has been done for you.

DO NOT WRITE IN ANY BARCODES.

There are **forty** questions on this paper. Answer **all** questions. For each question there are four possible answers **A, B, C** and **D**.

Choose the **one** you consider correct and record your choice in **soft pencil** on the separate Answer Sheet.

Read the instructions on the Answer Sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer.

Any rough working should be done in this booklet.

A copy of the Periodic Table is printed on page 16.

Electronic calculators may be used.

This document consists of **14** printed pages and **2** blank pages.

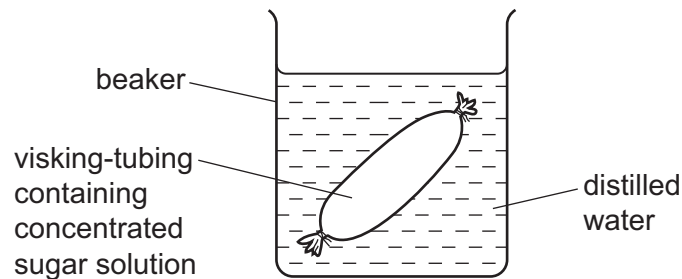


1 Which structures are found in both animal and plant cells?

- A cell membrane, cytoplasm and cell wall
- B chloroplasts, cytoplasm and cell wall
- C cytoplasm, cell membrane and nucleus
- D nucleus, cell wall and sap vacuole

2 Visking tubing is a partially permeable membrane.

Some visking tubing containing a concentrated sugar solution is weighed and placed in distilled water, as shown.



After 2 hours the visking tubing is removed and reweighed.

What happens to the mass and why?

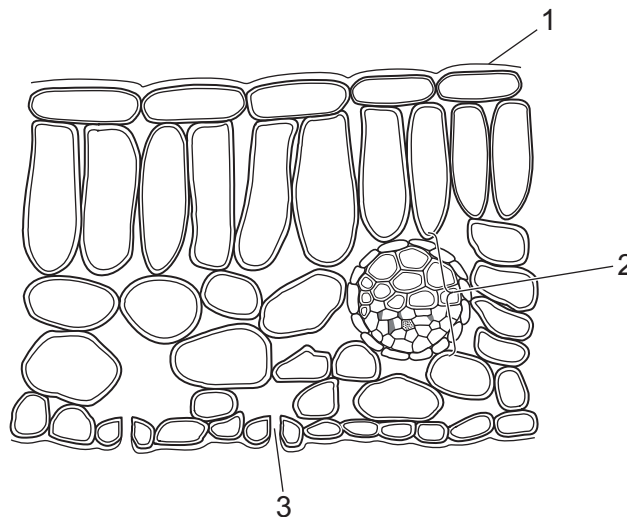
- A It decreases because sugar moves out.
 - B It decreases because water moves out.
 - C It increases because sugar moves in.
 - D It increases because water moves in.
- 3 Amylase is an enzyme important in seed germination.

What is the function of amylase in seed germination?

- A It breaks the testa so the plumule can emerge.
- B It causes the radical to elongate.
- C It changes the stored starch into sugars for respiration.
- D It helps the seed absorb water to rehydrate the cells.

3

4 The diagram shows a cross-section of a leaf.



What are the parts labelled 1, 2 and 3?

	1	2	3
A	cuticle	chloroplast	vascular bundle
B	cuticle	vascular bundle	stomata
C	mesophyll	chloroplast	vascular bundle
D	mesophyll	vascular bundle	stomata

5 What is the role of fat in the human body?

- A** to form glycogen
- B** to form urea
- C** to provide amino acids
- D** to provide a source of energy

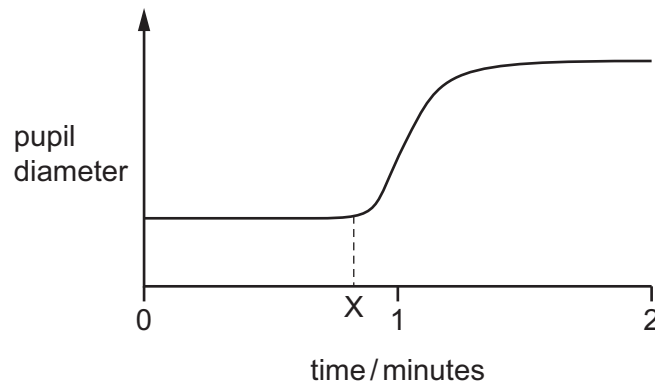
6 Which definition of transpiration is correct?

- A** loss of water by evaporation from the cuticle
- B** loss of water by evaporation from the stomata
- C** loss of water by osmosis from the cuticle
- D** loss of water by osmosis from the stomata

- 7 What is most likely to reduce the chance of suffering from coronary heart disease?
- A drinking more alcohol
 - B eating more animal fat
 - C giving up smoking
 - D taking less exercise
- 8 What is produced by anaerobic respiration in a muscle cell during exercise?
- A carbon dioxide and lactic acid
 - B carbon dioxide and water
 - C carbon dioxide only
 - D lactic acid only
- 9 Which row shows where carbon dioxide and urea are excreted from the body?

	carbon dioxide	urea
A	kidneys	kidneys
B	kidneys	liver
C	lungs	kidneys
D	lungs	liver

- 10 The graph shows how the diameter of the pupil of a person's eye changes during the course of two minutes.



What happens to the light intensity and the pupil diameter immediately after time X?

	light intensity	pupil diameter
A	decreases	decreases
B	decreases	increases
C	increases	decreases
D	increases	increases

- 11 Which part of the body is most likely to be directly damaged by drinking too much alcohol?

- A** the eyes
- B** the ileum
- C** the liver
- D** the lungs

- 12 Which human activity has caused most damage to tropical rain forests?

- A** burning fossil fuels
- B** flooding of land
- C** logging for timber
- D** searching for medicinal plants

- 13** A person suffers from pain when urinating. The cause of this symptom is a bacterial infection which can be treated using antibiotics.

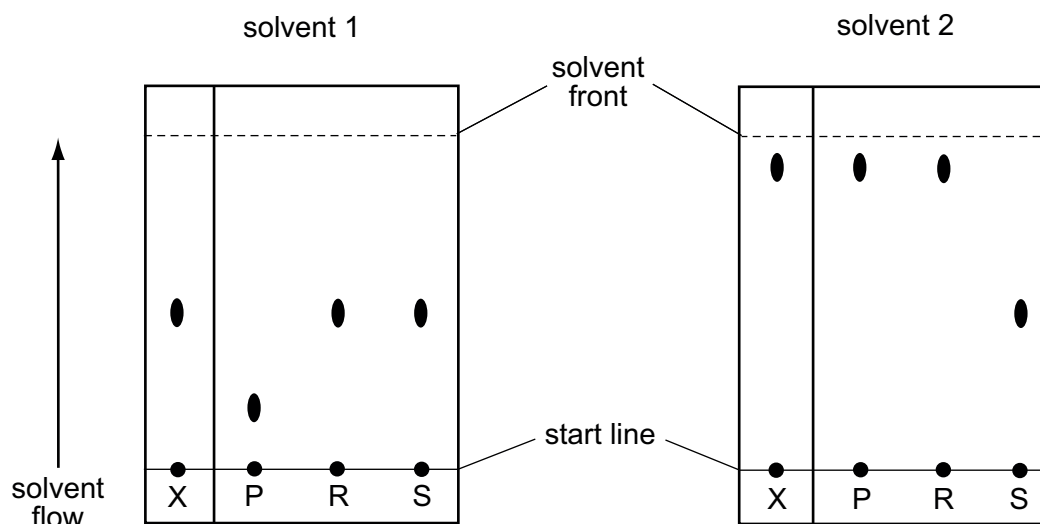
If left untreated, the disease can cause infertility or can be passed to an unborn child.

Which two diseases could both be the disease described?

- A** coronary heart disease and HIV
 - B** gonorrhoea and syphilis
 - C** HIV and gonorrhoea
 - D** syphilis and HIV
- 14** Solution X contains one or more of three substances P, R or S.

Two chromatograms, to compare X with each of the three substances, are obtained using different solvents.

The results are shown.



What does X contain?

- A** P only
 - B** R only
 - C** P and R
 - D** R and S
- 15** Which statement describes the particles in a liquid?
- A** They are separate from each other and move randomly in all directions.
 - B** They are separate from each other and vibrate forwards and backwards.
 - C** They touch each other and move randomly in all directions.
 - D** They touch each other and vibrate forwards and backwards.

16 Which statement about isotopes of the same element is correct?

- A They have different atomic numbers.
- B They have different chemical reactivities.
- C They have different nucleon numbers.
- D They have different numbers of electrons.

17 A particle has 10 electrons, 7 protons and 8 neutrons.

What is the symbol for the particle?

- A N^{3-} B O^{2-} C F^- D Ne

18 Which row describes most covalently bonded compounds?

	electrical conductivity when solid	melting point
A	conducts	high
B	conducts	low
C	insulator	high
D	insulator	low

19 What is the total number of atoms in $(\text{C}_2\text{H}_5)_2\text{O}$?

- A 3 B 9 C 13 D 15

20 The pH values of three solutions are shown.

	pH
ethanoic acid	6
hydrochloric acid	1
iron(III) chloride	3

What is the order of acidity of these solutions, from most acidic to least acidic?

- A ethanoic acid, hydrochloric acid, iron(III) chloride
- B ethanoic acid, iron(III) chloride, hydrochloric acid
- C hydrochloric acid, ethanoic acid, iron(III) chloride
- D hydrochloric acid, iron(III) chloride, ethanoic acid

21 Which statement about the elements in Group VII of the Periodic Table is **not** correct?

- A The melting points decrease down the group.
- B They are non-metals.
- C They exist as diatomic molecules.
- D They form ionic compounds with Group I elements.

22 Platinum is a metal.

Which statements about platinum are correct?

- 1 It can be hammered into shape.
- 2 It conducts heat.
- 3 It has a low boiling point.
- 4 It is shiny.
- 5 It is strong.

- A 1, 2, 3 and 4
- B 1, 2, 3 and 5
- C 1, 2, 4 and 5
- D 2, 3, 4 and 5

23 Four different metals are reacted separately with cold water, steam and dilute hydrochloric acid.

The results are shown.

metal	cold water	steam	dilute hydrochloric acid
W	no reaction	reacts slowly	reacts vigorously
X	no reaction	no reaction	reacts slowly
Y	reacts slowly	reacts vigorously	reacts explosively
Z	reacts slowly	reacts slowly	reacts vigorously

What is the order of reactivity of the four metals?

	least reactive → most reactive			
A	X	W	Z	Y
B	X	Z	W	Y
C	Y	W	Z	X
D	Y	Z	W	X

24 Polluted air is bubbled through distilled water at room temperature to form a solution.

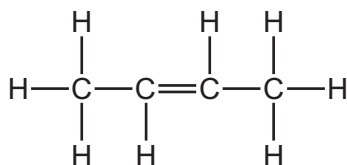
Solid sodium carbonate is added to the solution and bubbles of gas are produced.

Which pollutant could be present in the air?

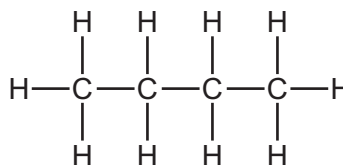
- A** a lead compound
- B** carbon dioxide
- C** carbon monoxide
- D** sulfur dioxide

25 Which structure does **not** represent C_4H_8 ?

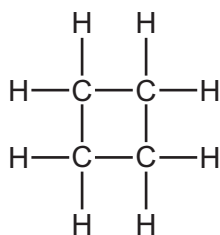
A



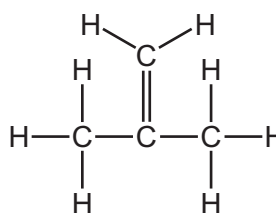
B



C



D



26 Which compound would **not** decolourise bromine water?

- A** C_2H_4
- B** C_2H_6
- C** C_3H_6
- D** C_4H_8

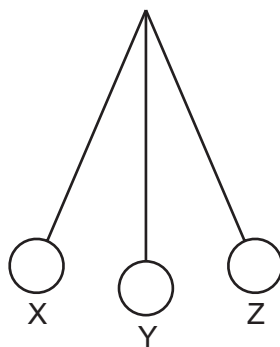
- 27 Two reactions are used to manufacture ethanol.

reaction 1 fermentation

reaction 2 addition of steam to ethene

Which statement is **not** correct?

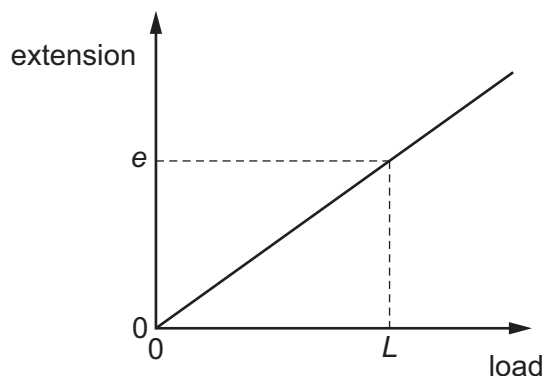
- A Reaction 1 requires a catalyst.
 - B Reaction 1 works at room temperature.
 - C Reaction 2 requires a catalyst.
 - D Reaction 2 works at room temperature.
- 28 A pendulum is repeatedly swinging from X through Y to Z and back again to X.
It takes 1 s to swing from X to Y.



How many periods of the pendulum are completed in 60 s?

- A 15
 - B 20
 - C 30
 - D 60
- 29 Which car, moving from rest, has an average acceleration of 2.0 m/s^2 ?
- A a car reaching a speed of 10 m/s in 2 s
 - B a car reaching a speed of 20 m/s in 5 s
 - C a car reaching a speed of 30 m/s in 10 s
 - D a car reaching a speed of 40 m/s in 20 s
- 30 What describes the density of a material?
- A the amount of matter in the material
 - B the mass per unit volume of the material
 - C the pull of gravity on the material
 - D the volume per unit mass of the material

- 31 The diagram shows an extension-load graph for an elastic object.



A load of L produces an extension of e .

What happens when the load L is removed?

- A The extension e continues to increase.
- B The extension e reduces but does not return to zero.
- C The extension e remains.
- D The extension e returns to zero.

- 32 Four people run up the same steps.

Which person produces the largest power?

	weight of person / N	time taken / s
A	300	4
B	400	5
C	500	10
D	600	15

- 33 On a cold day, a girl notices that the metal case of her mobile phone feels colder to touch than the glass screen.

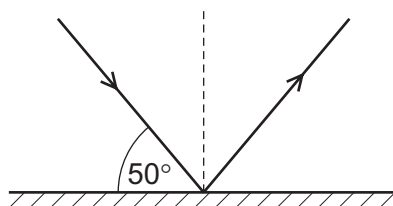
Which statement explains her observations?

- A Metal is a better conductor of heat than glass.
- B Metal is denser than glass.
- C Metal radiates heat less than glass.
- D The metal is thicker than the glass.

34 Which row correctly identifies examples of both a longitudinal and a transverse wave?

	longitudinal wave	transverse wave
A	light wave	radio wave
B	radio wave	sound wave
C	sound wave	surface water wave
D	surface water wave	light wave

35 The diagram shows a ray of light being reflected from a plane mirror.



The angle of incidence is increased by 10° .

What does the angle of reflection become?

- A** 30° **B** 40° **C** 50° **D** 60°

36 Which row gives correct units for current and voltage?

	current	voltage
A	C/s	J/C
B	C/s	J/s
C	Cs	J/C
D	Cs	J/s

37 A 12 V lamp uses a current of 2 A.

What is the resistance when the lamp is working correctly?

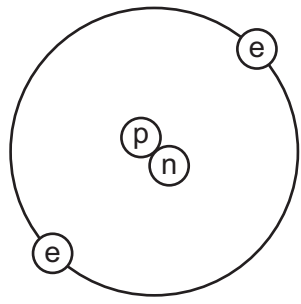
- A** $6\ \Omega$ **B** $10\ \Omega$ **C** $14\ \Omega$ **D** $24\ \Omega$

38 What is an example of induced magnetism?

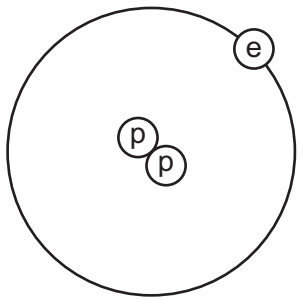
- A** a compass needle pointing north
B a north pole attracting iron filings
C a north pole repelling a north pole
D a negatively charged balloon attracting small pieces of paper

39 Which diagram represents a neutral atom?

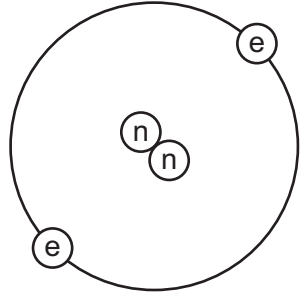
A



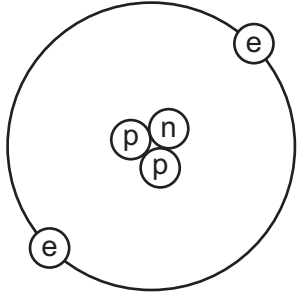
B



C



D



key

- (e) electron
- (n) neutron
- (p) proton

40 When using a sealed radioactive source, what is **not** a necessary safety precaution?

- A** checking the level of background radiation
- B** handling the source with long tongs
- C** keeping the exposure to a minimum
- D** using as weak a source as possible

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The Periodic Table of Elements

lanthanoids	57	La	lanthanum	139	58	Ce	cerium	140	59	Pr	praseodymium	141	60	Nd	neodymium	144	61	Pm	promethium	—	62	Sm	samarium	150	63	Eu	euroium	152	64	Gd	gadolinium	157	65	Tb	terbium	159	66	Dy	dysprosium	163	67	Ho	holmium	165	68	Er	erbium	167	69	Tm	thulium	169	70	Yb	ytterbium	173	71	Lu	lutetium	175
	89	Ac	actinium		90	Th	thorium	232	91	Pa	protactinium	231	92	U	uranium	238	93	Np	neptunium	—	94	Pu	plutonium	—	95	Am	americium	—	96	Cm	curium	—	97	Bk	berkelium	—	98	Cf	californium	—	99	Es	einsteinium	—	100	Fm	femium	—	101	Md	merdelvium	—	102	No	nobelium	—	103	Lr	lawrencium	—

The volume of one mole of any gas is 24 dm^3 at room temperature and pressure (r.t.p.).