



Cambridge International Examinations
Cambridge International General Certificate of Secondary Education

PHYSICAL SCIENCE

0652/62

Paper 6 Alternative to Practical

October/November 2016

MARK SCHEME

Maximum Mark: 60

Published

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Page 2	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
1(a)	1.10 g ;	1
1(b)	Only very slightly soluble in water / insoluble (in water) ;	1
1(c)(i)	225 ;	1
1(c)(ii)	0.9375 / 0.94 (accept 0.9) ;	1
1(d)(i)	filter funnel and filter paper shown and labelled ;	1
1(d)(ii)	0.16 ;	1
1(d)(iii)	$(1.10 - 0.16) = 0.94$;	1
1(e)(i)	$0.9375 / 1.10 \times 100 = 85 / 85.2$ (%) ; ecf from (a) and (c)(ii)	1
1(e)(ii)	$0.94 / 1.10 \times 100 = 85 / 85.5$ (%) ; ecf from (a) and (d)(iii)	1
1(f)	(Magnesium carbonate) reacts like calcium carbonate / also gives carbon dioxide / results in too much carbon dioxide / is not in the residue ;	1
	Total:	10

Question	Answers	Marks
2(a)(i)	copper / Cu ;	1
2(a)(ii)	23.0 ; 38.5 ;	2

Page 3	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
2(b)(i)	54(.0), 41.5, 30(.0), 15.5 ; ecf	1
2(b)(ii)	vertical scale linear and uses more than half of grid ; minimum of 3 points plotted correctly to within half a small square ; best straight line through origin ;	3
2(b)(iii)	agrees as all points close to / on straight line (through origin) ;	1
2(c)(i)	exothermic ;	1
2(c)(ii)	lid / insulation around cup / more accurate thermometer / repeat with different concentrations (extra points) ;	1
	Total:	10

Question	Answers	Marks
3(a)	note the reading on either side and find mean / shown on a diagram / measure cube and mark the <u>mid point</u> ;	1
3(b)(i)	36 (.0) ;	1
3(b)(ii)	21 (.0) cm ;	1
3(b)(iii)	14 (.0) cm ;	1
3(c)(i)	84.4 g ;	1
3(c)(ii)	56 to 56.2666... ; 56 / 56.3 g (2 / 3 significant figures) ;	2

Page 4	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
3(d)	any 2 centre of gravity of the rule not at the 50 cm mark ; difficulty in obtaining balance of ruler ; pivot not at right angles to edge of rule ; cube irregular ; ruler mass rounded ;	max 2
3(e)	<i>a</i> smaller and <i>b</i> greater ;	1
	Total:	10

Question	Answers	Marks
4(a)(i)	slit in a card (after lamp) ;	1
4(a)(ii)	light ray bends towards the normal as it enters water solution and meets P_w ; incident and refracted rays are parallel ;	2
4(a)(ii)	1.1 cm (± 0.1 cm) ;	1
4(a)(ii)	both angles of incidence and refraction correctly labelled ;	1
4(a)(ii)	2.0 cm (± 0.1 cm) ;	1
4(a)(ii)	the angle of refraction is greater when the light ray passes through glass /glass bends light more than water ;	1
4(b)(i)	reasonable accurate reflected ray drawn ; reflected angle marked between drawn normal and ray ;	2
4(b)(ii)	70° ;	1
	Total:	10

Page 5	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
5(a)(i)	ammonia / NH ₃ ;	1
5(a)(ii)	white precipitate / white solid ;	1
5(a)(iii)	add silver nitrate ; white precipitate ;	2
5(a)(iv)	blue ppt. ;	1
5(b)(i)	measuring cylinder / pipette / burette ;	1
5(b)(ii)	UI ; red to green ;	2
5(b)(iii)	salt would be impure / salt would be coloured ;	1
5(b)(iv)	heat ;	1
	Total:	10

Question	Answers	Marks
6(a)	64 (F) ; 49 (G) ;	2
6(b)	axes labelled with units ; suitable scales chosen for axes using at least half of grid ; at least 4 points plotted $\pm \frac{1}{2}$ square for each container ; smooth curves drawn (and labelled) ;	4
6(c)(i)	G is better (no mark) because the temperature fell more rapidly / lower curve ;	1

Page 6	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
6(c)(ii)	named non-metallic material for F , e.g. polythene / wood / plastic ; named metallic material for G , e.g. copper / metal ;	2
6(d)	value in region 21 to 40 °C ;	1
	Total:	10