UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

MARK SCHEME for the May/June 2010 question paper for the guidance of teachers

5129 COMBINED SCIENCE

5129/02

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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Page 2	Mark Scheme: Teachers' version	Syllabus	Paper
	GCE O LEVEL – May/June 2010	5129	02

1	(a)	correct symbol in parallel with 6Ω resistor	[2]
	(b)	(i) V = IR (or equivalent) or 0.2 × 6 = 1.2	
		V (unit independent)	[3]
		(ii) 9	[1]
2	(a)	magnesium and aluminium / Mg and A <i>l</i> metals	[2]
		(answers are independent)	[-]
	(b)	blue / purple / violet	[0]
		red / orange / yellow	[2]
	(c)	same number of / 2 electrons in outermost shell (allow valence electrons)	[1]
3	(a)	(i) 0 and 1	[1]
		(ii) 2 and 3	[1]
		(iii) 3 and 4	[1]
	(b)	iris / circular and/or radial muscles	[1]
	(c)	a line which remains above the drawn line throughout (can be along initial line but must not increase)	[1]
	(d)	retina	[1]
4	(a)	27	[1]
	(b)	electron / e	[1]
	(c)	32	[1]
	(d)	an indication of 2 half lives = 11 400	[2]
		2 × 5700 with incorrect answer = 1 mark	

Page 3		ge 3			hers' version	Syllabus	Paper
			GCE O LEVE	EL – Ma	y/June 2010	5129	02
5		liver amino ac kidneys liver	cids				[4]
6	(a)	neutralis	eation				[1]
	(b)	17 80					[2]
	(c)	(17 × 2) / = 0.425 k (ecf from	kg / 425 g				[2]
7	(a)		ack is a better emitter / ı / white is a poorer emitt				[1]
	(b)	conduction	on				[1]
	(c)		on air expands air is less dense	}	any one		[1]
8	(a)	filter add chlo sedimen		}	any two		[2]
	(b)	kill bacte	solids / insoluble particle eria / micro-organisms ettle to the bottom	es / dirt			[2]
9	(a)		sade / mesphyll er) epidermis				[2]
	(b)	spongy r	<u>mesophyll</u>				[1]
	(c)	by diffusi through s					[2]
	(d)		over short distance / ea etrates to all parts / mo				[2]

Page 4	Mark Scheme: Teachers' version	Syllabus	Paper
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10	(a)	no yes no yes 4 = 2 marks 3 = 1 mark	[2]
	(b)	(i) plastic	[1]
		(ii) iron / Fe	[1]
11	(a)	haematite / magnetite	[1]
	(b)	(i) 3 2 3 (allow correct multiples)	[1]
		(ii) removal of oxygen / gain of electrons / lower oxidation state	[1]
		(iii) coke / carbon burns to form carbon dioxide carbon dioxide reacts with carbon carbon reacts with oxygen = 1 mark	[2]
	(c)	too reactive / more reactive than iron / carbon	[1]
12	(a)	(i) 60	
		(ii) 8 (ignore any sign)	[2]
	(b)	oscillation / disturbance / displacement / motion of particles perpendicular to the direction / motion of the wave	[2]
13	(a)	break down of large pieces to small pieces / crush / tear / grind food increase surface area mix food with saliva / enzymes make swallowing easier (ignore references to chewing / digestion) any two	[2]
	(b)	B (correctly identified) fewer decayed teeth / less decay	[1]
	(c)	Town B: eat less sugar have better dental care / more clinics / dentists brush teeth more visit dentist more regularly use more mouthwash eat more fruit and vegetables better education	[2]

accept converse in terms of town $\ensuremath{\mathsf{A}}$

[1]

Paper

Syllabus

	ı ağc c		mark contine. Teachers version	Cynabas	i apci
			GCE O LEVEL – May/June 2010	5129	02
14	(a)	infra red	/ ir		[1]
	(b)	radio			[1]
	(c)	300 000	000 / 3 × 10 ⁸		[1]
15	(a)	carbon d	ioxide		
		plan	ts animals		[3]
	(b)	X = photo Y = respi	osynthesis		
			y / decomposition / (bacterial) respiration		[3]
16	(a)	D			[1]
	(b)	В			[1]
	(c)	С			[1]
17	(a)	time = di 4.0	stance/speed or 0.8/0.2		[2]
	(b)	work = fo	orce × distance or 4 × 0.8		
		J / joules	(unit independent)		[3]
18	(a)	carbon d	ioxide team (any order)		[2]
					[-]
	(b)	carbon n poisonou	nonoxide us / toxic / correct description of mode of action		[2]
	(c)		neral formula		[1]

Mark Scheme: Teachers' version

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any one

similar **chemical** properties

formulae differ by CH₂

gradation in physical properties

Page 6	Mark Scheme: Teachers' version	Syllabus	Paper
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19	(a)	potential / gravitational kinetic / movement	[2]
	(b)	0.5	[1]
20	(a)	(i) any date from Feb 28 to Mar 5(ii) any date from Mar 10 to Mar 17	[1] [1]
	(b)	 (i) 28 days / 4 weeks (ii) age / menopause stress / emotional state / anxiety diet / malnutrition / starvation pregnancy / breast feeding exercise genetic factors / inheritance 	[1]
21	(a)	fermentation / anaerobic respiration	[1]
	(b)	(i) provides enzymes	[1]
		(ii) prevents oxidation of ethanol / formation of ethanoic acid	[1]
	(c)	(fractional) distillation	[1]
	(d)	correct structure of ethanol	[1]