



**Cambridge International Examinations**  
Cambridge International General Certificate of Secondary Education

---

**COMBINED SCIENCE**

**0653/63**

Paper 6 Alternative to Practical

**May/June 2016**

MARK SCHEME

Maximum Mark: 60

---

**Published**

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2016 series for most Cambridge IGCSE<sup>®</sup>, Cambridge International A and AS Level components and some Cambridge O Level components.

Page 2	Mark Scheme	Syllabus	Paper
	Cambridge IGCSE – May/June 2016	0653	63

- 1 (a) incisor (front) ;  
molar (back) ; [2]
- (b) (i) dissolve sample of plaque in distilled water ;  
use of full range indicator/pH meter ; [2]
- (ii) below 7 ; [1]
- (iii) acid produced by bacteria/sugar forms acid ; [1]
- (c) 2 groups – brushing twice and brushing three times ;  
over several days/weeks ;  
compare/measure amount of staining ;  
less staining means less plaque ;  
swap groups over as a control ; [max 4]

[Total: 10]

- 2 (a) (i) salt **C** label pointing to residue in filter paper  
**AND**  
salt **B** label pointing to filtrate in beaker ; [1]
- (ii) correct residue label  
**AND**  
correct filtrate label ; [1]

(b) (i)

	conclusion
(add HCl)	not carbonate / not $\text{CO}_3^{2-}$ ;
(add HCl + BaCl)	sulfate / $\text{SO}_4^{2-}$ ;
(add NaOH)	copper(II) / $\text{Cu}^{2+}$ ;

[3]

- (ii) copper(II) sulfate ; [1]

- (c) (i) limewater goes milky/white ppt. ; [1]
- (ii) white ppt. ;  
ppt. dissolves ; [2]
- (iii)  $\text{ZnCO}_3$  ; [1]

[Total: 10]

Page 3	Mark Scheme	Syllabus	Paper
	Cambridge IGCSE – May/June 2016	0653	63

- 3 (a) 0.19 (V) ; [1]
- (b)  $R$  values correct (should be: 0.79, 2.42, 4.00) ;  
consistent significant figures ; [2]
- (c) axes labelled with units ;  
suitable choice of scales ( $\geq \frac{1}{2}$  the grid used) ;  
plots correct to  $\frac{1}{2}$  small square ;  
good best-fit line judgement ; [4]
- (d) directly proportional ;  
straight line through the origin ; [2]
- (e) switch off between readings / fan the wire / resistor in series with the wire ; [max 1]
- [Total: 10]**
- 4 (a) good size drawing with clear lines ;  
correct shape ; [2]
- (b) (i) correct measurement (34 mm) ; [1]
- (ii) correct measurement (from their drawing) ; [1]
- (iii) correct calculation ; [1]
- (c) (i) 3 correct labels ;;; [3]
- (ii) (agree)  
cell wall and nucleus = 1 mark ;  
*any one from:* starch grain / vacuole / chloroplast ; [max 2]
- [Total: 10]**
- 5 (a) (i) 71.8 ;  
79.6 ; [2]
- (ii) 20.3 ;  
28.1 ; [2]
- (iii) 48 ; [1]
- (iv) not all iron reacted / not hot enough ; etc. [1]
- (b) chlorine / gas is toxic ; [1]

Page 4	Mark Scheme	Syllabus	Paper
	Cambridge IGCSE – May/June 2016	0653	63

- (c) use of sodium hydroxide ;  
 mention of dissolving, water, solution or aqueous ;  
 iron(II) green ppt. **AND** iron(III) brown ppt.; [3]

[Total: 10]

- 6 (a) (i) 118 ; [1]

- (ii) 83 (only) ; [1]

- (iii) max use of paper e.g. vertical axis starts at 30 ;  
 correct plotting ;  
 smooth curves ;  
 (at least) one curve labelled ; [4]

- (iv) (*similar*) both start at same temp/ both go down, etc. ;  
 (*different*) go down at different rates/ end at different temps, etc. ; [2]

- (b) result at 8 mins is wrong/ anomalous ; [1]

- (c) e.g. pour same volume of water into each container ;  
 or record initial temperatures in the beakers ; [max 1]

[Total: 10]