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General Certificate of Education

2015

Centre Number

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Candidate Number

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Physics

Assessment Unit A2 1

assessing

Momentum, Thermal Physics,
Circular Motion, Oscillations
and Atomic and Nuclear Physics



AY211

[AY211]

TUESDAY 19 MAY, MORNING

TIME

1 hour 30 minutes.

INSTRUCTIONS TO CANDIDATES

Write your Centre Number and Candidate Number in the spaces provided at the top of this page.

Answer **all seven** questions.

Write your answers in the spaces provided in this question paper.

INFORMATION FOR CANDIDATES

The total mark for this paper is 90.

Quality of written communication will be assessed in Question **3(a)**.

Figures in brackets printed down the right-hand side of pages indicate the marks awarded to each question.

Your attention is drawn to the Data and Formulae Sheet which is inside this question paper.

You may use an electronic calculator.

Question **7** contributes to the synoptic assessment required of the specification.

For Examiner's
use only

Question Number	Marks	Remark
1		
2		
3		
4		
5		
6		
7		

Total
Marks

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[2]

9432

- 2 (a) State the ideal gas equation, identify the terms and use it to explain why putting an aerosol container onto a fire could cause it to explode.

[5]

- (b) A cylindrical can containing a gas at standard temperature and pressure has diameter of 5.0 cm and height 15.0 cm, as shown in Fig. 2.1.

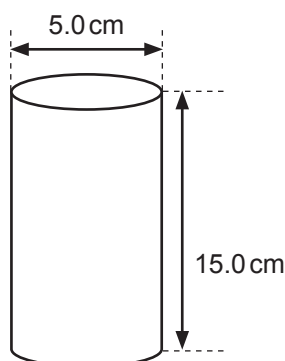


Fig. 2.1

- (i) One mole of an ideal gas occupies a volume of $22.4 \times 10^{-3} \text{ m}^3$ at standard temperature and pressure. Calculate the number of moles of the gas inside the can, assuming the gas is ideal.

Number of moles = _____

[3]

Examiner Only	
Marks	Remark

- (b) In a system undergoing simple harmonic motion, at a time $t = 0.6\text{ s}$, the acceleration, a is -0.74 m s^{-2} and the displacement, x is 0.3 m . Calculate the period and the amplitude of the motion and use your values to draw an accurate graph for the motion of the object on the grid of **Fig. 3.2**. At time $t = 0$, the system is at maximum displacement. On your graph show at least 2 complete cycles of the motion.

Period = _____ s

Amplitude = _____ m

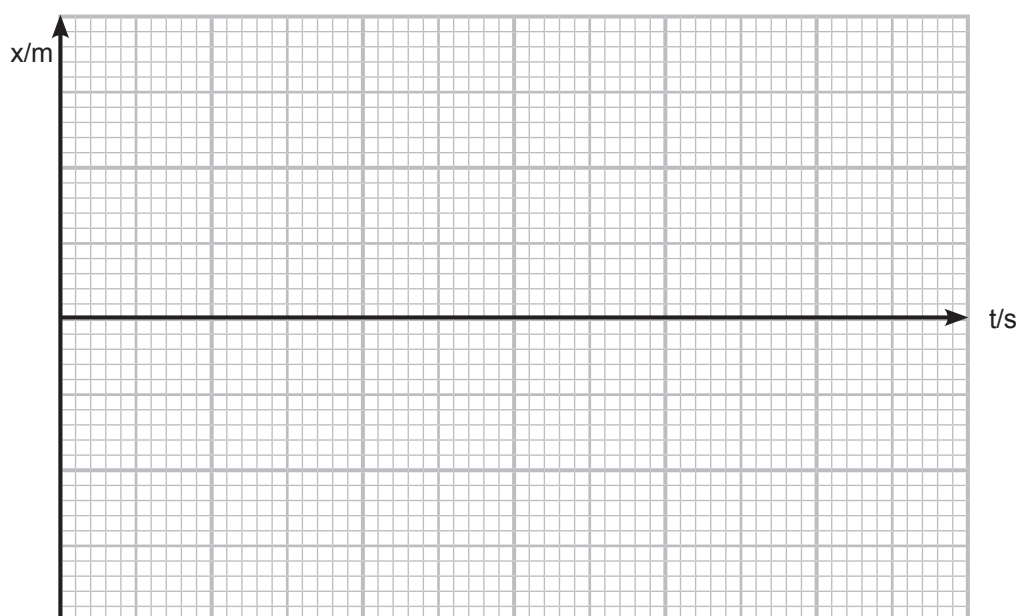


Fig. 3.2

[7]

- (c) State one similarity and one difference between a system that is critically damped and one that is overdamped.

[2]

[2]

- (b)** A more recent use of alpha particles is in cancer treatments. One example is the alpha decay of radium-223 which can be used to kill cancer cells on bone surfaces. Radium-223 has a half-life of 11.4 days.

A sample of radium with an activity of 93 kBq per kg of body mass is required for a particular patient. The patient has a mass of 76 kg and is due for treatment at 9 a.m. Calculate the initial activity required if the sample is prepared at 3 p.m. the previous day.

Initial activity = _____ Bq [4]

Examiner Only	
Marks	Remark

- (c) It is important that the half-life of the isotope is found accurately so that the previous calculation does not result in an incorrect dose being received by the patient.

One method used to determine the half-life of a radioactive isotope is to measure the activity over a period of time and plot a graph of activity against time. The time taken for the activity to halve can be read directly from the graph for more than one value and the results averaged.

- (i) **Fig. 4.1** shows an activity against time graph for a sample of radium-223. Scale the time axis of **Fig. 4.1**. [1]

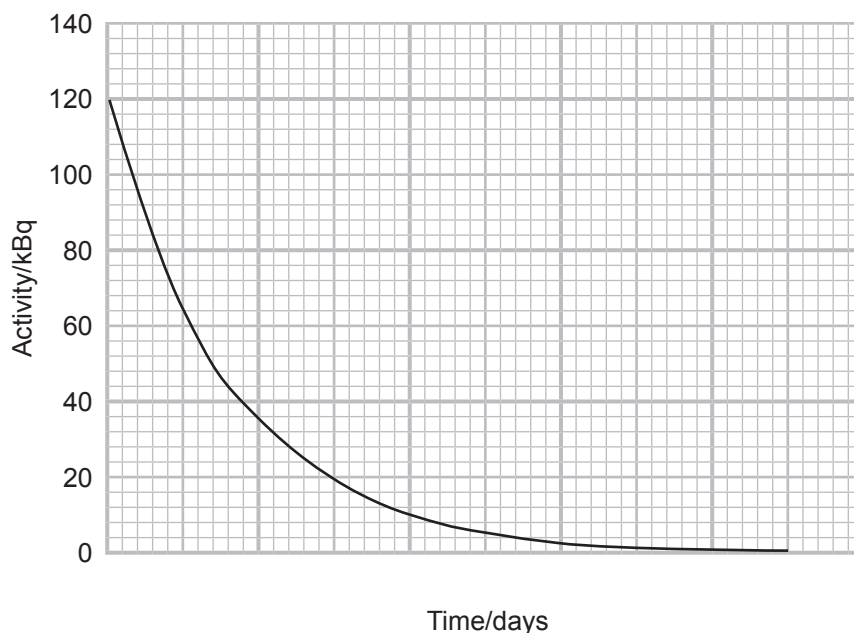


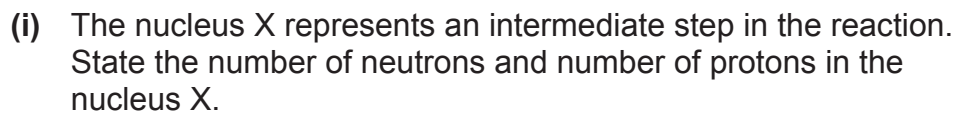
Fig. 4.1

- (ii) Describe an alternative graphical method that could be used to find a value for the half-life, given a series of readings of activity and time.

[3]

Examiner Only	
Marks	Remark

(b) Equation 5.1 shows one example of a fission reaction.



Number of protons = _____ [2]

Table 5.1

Particle	Rest mass/u
U-235	235.0439
Cs-137	136.9070
Rb-95	94.9290
n	1.0087

Calculate the energy released in this fission reaction in eV.

Energy released = _____ eV [5]

Examiner Only	
Marks	Remark

[2]

(ii) Explain why these benefits are described as ‘potential’. Other than plasma confinement, state one difficulty that needs to be overcome to change this.

[2]

(b) (i) The JET reactor uses magnetic confinement to contain the plasma formed as part of the fusion process. What property do the particles of the plasma have that allows them to be controlled by a magnetic field?

[1]

(ii) State the shape of the vessel in which the plasma is contained in the JET reactor.

[1]

Examiner Only	
Marks	Remark

- 7 An object was heated to a temperature of 90°C and left to cool in an area with a controlled, constant temperature of 0°C . The temperature, θ , of the object was recorded over a time period of 150 seconds and the results plotted on the grid of **Fig. 7.1**.

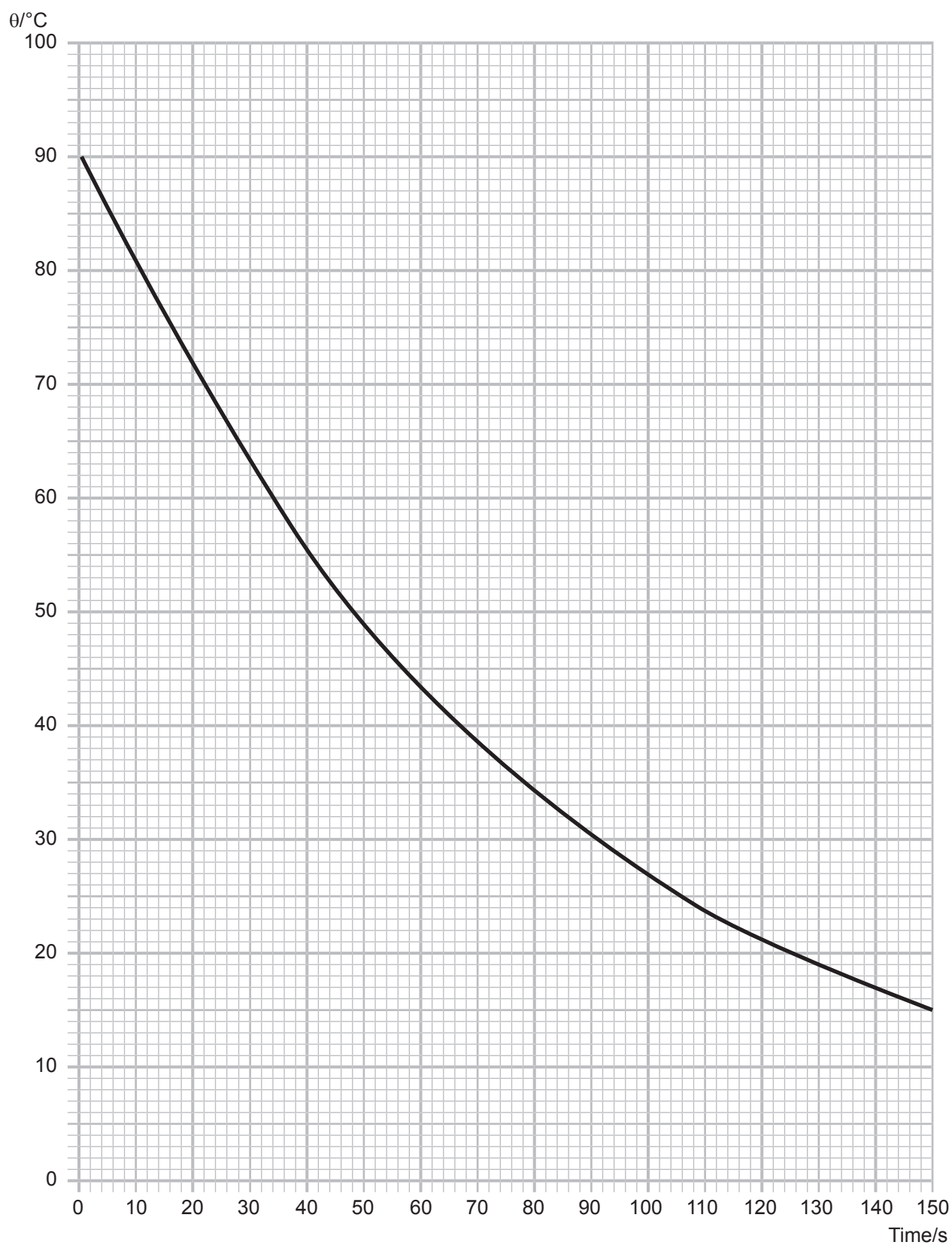


Fig. 7.1

- (b) **Equation 7.1** describes the graph shown in **Fig. 7.1**. On the axis of **Fig. 7.3** sketch a graph of $\ln \theta$ against time, t . Include values for intercepts on the x and y axes.

$$\theta = 90e^{-0.012t}$$

Equation 7.1



Fig. 7.3

[4]

THIS IS THE END OF THE QUESTION PAPER

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GCE Physics

Data and Formulae Sheet for A2 1 and A2 2

Values of constants

speed of light in a vacuum	$c = 3.00 \times 10^8 \text{ m s}^{-1}$
permittivity of a vacuum	$\epsilon_0 = 8.85 \times 10^{-12} \text{ F m}^{-1}$ $\left(\frac{1}{4\pi\epsilon_0} = 8.99 \times 10^9 \text{ F}^{-1} \text{ m} \right)$
elementary charge	$e = 1.60 \times 10^{-19} \text{ C}$
the Planck constant	$h = 6.63 \times 10^{-34} \text{ J s}$
(unified) atomic mass unit	$1 \text{ u} = 1.66 \times 10^{-27} \text{ kg}$
mass of electron	$m_e = 9.11 \times 10^{-31} \text{ kg}$
mass of proton	$m_p = 1.67 \times 10^{-27} \text{ kg}$
molar gas constant	$R = 8.31 \text{ J K}^{-1} \text{ mol}^{-1}$
the Avogadro constant	$N_A = 6.02 \times 10^{23} \text{ mol}^{-1}$
the Boltzmann constant	$k = 1.38 \times 10^{-23} \text{ J K}^{-1}$
gravitational constant	$G = 6.67 \times 10^{-11} \text{ N m}^2 \text{ kg}^{-2}$
acceleration of free fall on the Earth's surface	$g = 9.81 \text{ m s}^{-2}$
electron volt	$1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$

The following equations may be useful in answering some of the questions in the examination:

Mechanics

Conservation of energy $\frac{1}{2}mv^2 - \frac{1}{2}mu^2 = Fs$ for a constant force

Hooke's Law $F = kx$ (spring constant k)

Simple harmonic motion

Displacement $x = A \cos \omega t$

Sound

Sound intensity level/dB $= 10 \lg_{10} \frac{I}{I_0}$

Waves

Two-source interference $\lambda = \frac{ay}{d}$

Thermal physics

Average kinetic energy of a molecule $\frac{1}{2}m\langle c^2 \rangle = \frac{3}{2}kT$

Kinetic theory $pV = \frac{1}{3}Nm\langle c^2 \rangle$

Thermal energy $Q = mc\Delta\theta$

Capacitors

Capacitors in series $\frac{1}{C} = \frac{1}{C_1} + \frac{1}{C_2} + \frac{1}{C_3}$

Capacitors in parallel $C = C_1 + C_2 + C_3$

Time constant $\tau = RC$

Light

Lens formula

$$\frac{1}{u} + \frac{1}{v} = \frac{1}{f}$$

Magnification

$$m = \frac{v}{u}$$

Electricity

Terminal potential difference

$$V = E - Ir \quad (\text{e.m.f. } E; \text{ Internal Resistance } r)$$

Potential divider

$$V_{\text{out}} = \frac{R_1 V_{\text{in}}}{R_1 + R_2}$$

Particles and photons

Radioactive decay

$$A = \lambda N$$

$$A = A_0 e^{-\lambda t}$$

Half-life

$$t_{\frac{1}{2}} = \frac{0.693}{\lambda}$$

de Broglie equation

$$\lambda = \frac{h}{p}$$

The nucleus

Nuclear radius

$$r = r_0 A^{\frac{1}{3}}$$

