



GCSE (9-1)

Combined Science B (Twenty First Century)

Unit **J260/08**: Combined Science

General Certificate of Secondary Education

Mark Scheme for June 2018

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This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which marks were awarded by examiners. It does not indicate the details of the discussions which took place at an examiners' meeting before marking commenced.

All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

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Annotations available in RM Assessor

Annotation	Meaning
	Correct response
	Incorrect response
	Omission mark
	Benefit of doubt given
	Contradiction
	Rounding error
	Error in number of significant figures
	Error carried forward
	Level 1
	Level 2
	Level 3
	Benefit of doubt not given
	Noted but no credit given
	Ignore

Abbreviations, annotations and conventions used in the detailed Mark Scheme (to include abbreviations and subject-specific conventions).

Annotation	Meaning
/	alternative and acceptable answers for the same marking point
✓	Separates marking points
DO NOT ALLOW	Answers which are not worthy of credit
IGNORE	Statements which are irrelevant
ALLOW	Answers that can be accepted
()	Words which are not essential to gain credit
—	Underlined words must be present in answer to score a mark
ECF	Error carried forward
AW	Alternative wording
ORA	Or reverse argument

Subject-specific Marking Instructions**INTRODUCTION**

Your first task as an Examiner is to become thoroughly familiar with the material on which the examination depends. This material includes:

- the specification, especially the assessment objectives
- the question paper
- the mark scheme.

You should ensure that you have copies of these materials.

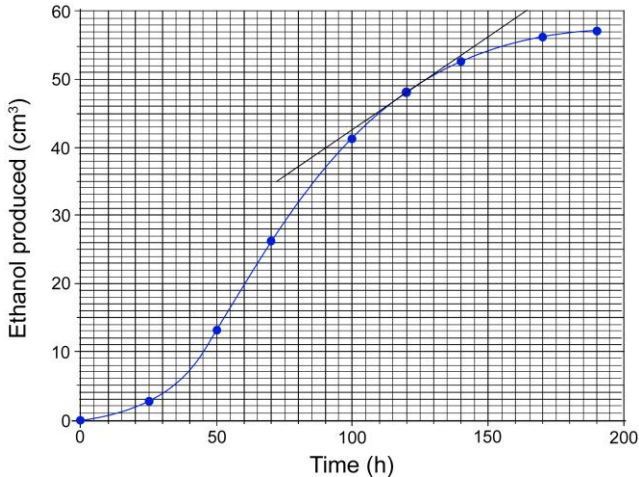
You should ensure also that you are familiar with the administrative procedures related to the marking process. These are set out in the OCR booklet **Instructions for Examiners**. If you are examining for the first time, please read carefully **Appendix 5 Introduction to Script Marking: Notes for New Examiners**.

Please ask for help or guidance whenever you need it. Your first point of contact is your Team Leader.

The breakdown of Assessment Objectives for GCSE (9-1) in Combined Science B:

	Assessment Objective
AO1	Demonstrate knowledge and understanding of scientific ideas and scientific techniques and procedures.
AO1.1	Demonstrate knowledge and understanding of scientific ideas.
AO1.2	Demonstrate knowledge and understanding of scientific techniques and procedures.
AO2	Apply knowledge and understanding of scientific ideas and scientific enquiry, techniques and procedures.
AO2.1	Apply knowledge and understanding of scientific ideas.
AO2.2	Apply knowledge and understanding of scientific enquiry, techniques and procedures.
AO3	Analyse information and ideas to interpret and evaluate, make judgements and draw conclusions and develop and improve experimental procedures.
AO3.1	Analyse information and ideas to interpret and evaluate.
AO3.1a	Analyse information and ideas to interpret.
AO3.1b	Analyse information and ideas to evaluate.
AO3.2	Analyse information and ideas to make judgements and draw conclusions.
AO3.2a	Analyse information and ideas to make judgements.
AO3.2b	Analyse information and ideas to draw conclusions.
AO3.3	Analyse information and ideas to develop and improve experimental procedures.
AO3.3a	Analyse information and ideas to develop experimental procedures.
AO3.3b	Analyse information and ideas to improve experimental procedures.

Question		Answer	Marks	AO element	Guidance
	(ii)	Comparison of e-cigarette is a physical change and cigarette is a chemical change ✓	1	2.1	ALLOW new products formed (including gases) in cigarettes
(d)	(i)	Ali ✓	1	3.1b	
	(ii)	Sarah ✓	1	3.1b	
(e)*		<p><i>Please refer to the marking instructions on page 5 of this mark scheme for guidance on how to mark this question.</i></p> <p>Level 3 (5–6 marks) Analyses data to form reasoned conclusions about the relative risk and presence or lack of correlation.</p> <p><i>There is a well-developed line of reasoning which is clear and logically structured. The information presented is relevant and substantiated.</i></p> <p>Level 2 (3–4 marks) Analyses some data to form conclusions about the risk and presence or lack of correlation.</p> <p><i>There is a line of reasoning presented with some structure. The information presented is relevant and supported by some evidence.</i></p> <p>Level 1 (1–2 marks) Identifies foods from the data that change the risk of Parkinson's disease.</p> <p><i>There is an attempt at a logical structure with a line of reasoning. The information is in the most part relevant.</i></p> <p>0 marks <i>No response or no response worthy of credit.</i></p>	6	3 × 3.1a 3 × 3.2b	<p>AO3.1a Analyse data For example:</p> <ul style="list-style-type: none"> • reduction of risk linked with eating all foods except tomato juice • peppers - 0.24 reduced risk (conc. 102) • tomatoes – 0.58 reduced risk (conc. 44) • potatoes – 0.92 reduced risk (conc. 19) • tomato juice – 2.16 increases risk (conc. 30) <p>AO3.2b Analyse information to make conclusions/correlations</p> <ul style="list-style-type: none"> • Idea that results from tomato juice suggest that other factors may be involved. • Correlations imply that nicotine-containing foods give protection against Parkinson's disease • Portion may alter risk • Comparative statements about risk • Correlation ideas limited by small sample size • other factors may be involved in patients who ate nicotine-containing foods

Question	Answer	Marks	AO element	Guidance
3 (a)	2, 2 ✓	1	2.1	Both must be correct for mark.
(b) (i)	Any one from: Colour change / colorimetry ✓ Formation of a precipitate ✓ Change in temperature / mass / volume of gas ✓	1	1.2	
(ii)	 <p>Draw tangent (at or over 120 h) ✓</p> <p>Select two suitable points to calculate gradient from the tangent ✓</p> <p>Interpolate values of x and y at these two points ✓</p> <p>Uses formula to calculate rate ✓</p> <p>Correct answer given ✓</p>	5	1.2 2.2 2.2 1.2 2.2	<p>Must show tick or cross on graph</p> <p>ALLOW 0.129 / 0.13 for 5 marks</p> <p>ALLOW +/- ½ small square</p> <p>If no tangent drawn only award last 2 marking points</p> <p>For example:</p> $\text{rate of reaction} = \frac{\text{change in amount}}{\text{change in time}}$ $= \frac{59 - 36}{160 - 75} = \frac{23}{85} = 0.27 \text{ cm}^3 / \text{h}$

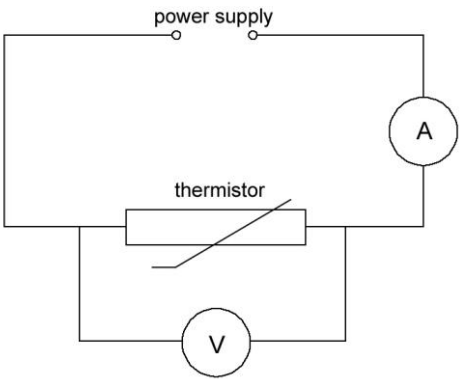
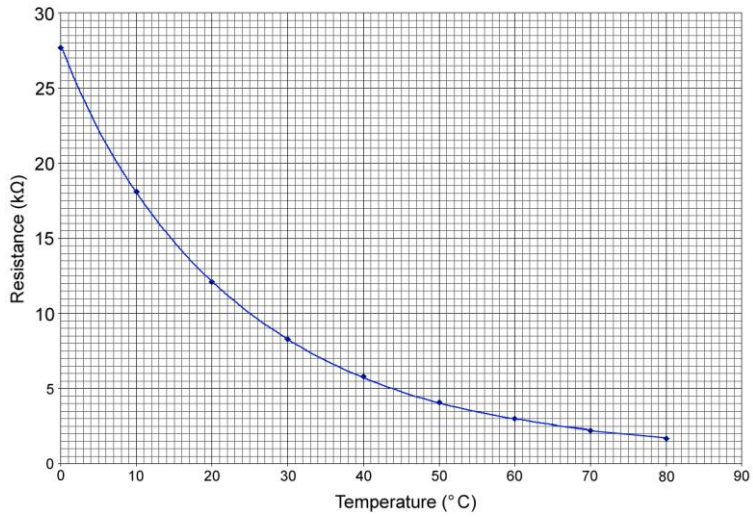
Question			Answer	Marks	AO element	Guidance
	(c)	(i)	<p>Any one from: Increasing temperature increases rate up to the optimum / decreases the rate after the optimum temperature ✓</p> <p>Optimum temperature = 32.5°C ✓</p> <p>Any two from: After 32.5°C enzymes in yeast denature ✓</p> <p>Activity of enzyme drops after optimum temperature ✓</p> <p>Active site in enzyme changes so that the reactant no longer fits ✓</p> <p>Uses ideas of collision theory to explain the effect of temperature on the rate of reaction ✓</p>	3	<p>3.1a</p> <p>3.2b x2</p>	<p>ALLOW 32-33 for optimum temperature (½ small square)</p>
		(ii)	<p>Any two from: Make measurements at smaller intervals of temperature; ✓</p> <p>Specified intervals, e.g. 1 °C/0.5 °C ✓</p> <p>Identifies a suitable temperature range / around the optimum temperature ✓</p>	2	2 x 3.3b	<p>ALLOW any interval less than 5°C</p> <p>Example of a suitable range might be 30 °C to 35 °C</p>

Question		Answer	Marks	AO element	Guidance
	(d)	<p>Any two from:</p> <p>Starting materials for fermentation renewable / sustainable / more readily available / ORA ✓</p> <p>ethene is also important as chemical feedstock ✓</p> <p>Less energy /energy costs are lower for fermentation ✓</p> <p>Fermentation uses lower temperature/ energy requirement / ORA ✓</p> <p>Fermentation uses lower pressure / the industrial production uses high pressure ✓</p> <p>Reaction of ethene reversible / reaction comes to equilibrium ✓</p>	2	2 × 2.1	<p>Take 'it' as referring to the fermentation reaction</p> <p>ALLOW Safety argument qualified e.g. safer as no phosphoric acid is used.</p>

Question			Answer	Marks	AO element	Guidance
4	(a)	(i)	F ✓ T ✓ T ✓	3	2.1 3.1a 3.1a	
		(ii)	OPEN-LOOP Recycling Any four from: Sort PET waste ✓ Grind to flake / shredded / cut into pieces ✓ Separate contaminants ✓ Washed / clean ✓ Dried ✓ Melted (into pellets) ✓ (Flake) converted directly to useful products ✓ OR CLOSED-LOOP Recycling Any four from: Sort PET waste ✓ Clean PET waste ✓ converted to monomers / depolymerised ✓ Repolymerise the pure monomer/ new PET synthesised ✓ PET/ polymer granules made into new (food-grade) products ✓	4	4 × 1.1	ALLOW named useful products e.g. carpets, non-food containers, filling for pillow, quilts and jackets.

Question			Answer	Marks	AO element	Guidance
	(b)	(i)	<p>Any one from:</p> <p>Less energy costs ✓</p> <p>Easy to culture bacteria (so more bottles could be broken down) ✓</p>	1	2.1	
		(ii)	<p>Any three from:</p> <p>Bacteria didn't need to adapt before 1973 ✓</p> <p>Bacteria mutated (after 1973) ✓</p> <p>Mutation allows the bacteria to break down PET ✓</p> <p>By producing an enzyme / mutation allows bacteria to use PET as a food source ✓</p> <p>Long time for bacteria to evolve / mutation to be observed (43 years) ✓</p> <p>Bacteria have selective advantage (over those without mutant gene allele) ✓</p> <p>PLUS</p> <p>Gene / allele / variant spreads throughout population ✓</p>	4	<p>3 × 3.1b</p> <p>2.1</p>	

Question			Answer	Marks	AO element	Guidance
5	(a)	(i)	1 mark for beta ✓ 2 marks for nitrogen isotope ✓✓ ${}^{14}_6\text{C} \longrightarrow {}^{14}_7\text{N} + {}^0_{-1}\beta$	3	1.1 2 x 2.1	
		(ii)	5750 years ✓ Evidence of at least one correct half-life construction on the graph ✓	2	1.2 3.1b	ACCEPT answers to ± 250 years Must indicate a tick or cross on graph
	(b)	(i)	Concentration of ${}^{14}\text{C}$ in the environment has remained constant ✓	1	2.1	ALLOW idea that no more carbon(-14) enters the body after death
		(ii)	Half-life of ${}^{14}\text{C}$ is short ✓ Half-life of ${}^{40}\text{K}$ is long ✓ Any three from: ${}^{14}\text{C}$ has little activity after millions of years / small amounts of ${}^{14}\text{C}$ left in the fossil ✓ Idea that ${}^{14}\text{C}$ decays too quickly (to date fossils) ✓ Idea that ${}^{40}\text{K}$ decays slowly enough (to date much older objects / fossils from millions of years ago) ✓ (${}^{40}\text{K}$ present is large enough) to give accurate estimation of age of the fossils ✓ All ${}^{40}\text{Ar}$ must be derived from ${}^{40}\text{K}$ ✓	5	2 x 3.1a 3 x 3.2b	ALLOW Comparative statement of ${}^{40}\text{K}$ half-life longer than ${}^{14}\text{C}$ for 2 marks ALLOW almost all ${}^{14}\text{C}$ has decayed after 45,000 years (from graph) ALLOW ${}^{14}\text{C}$ decays more quickly / ORA for 2 marks
	(c)		In (the king's) diet/as food ✓	1	3.2a	ALLOW breathed in / from air

Question	Answer	Marks	AO element	Guidance
6 (a) (i)	<p>Ammeter in series and voltmeter in parallel ✓ Correct symbol for thermistor ✓</p> 	2	2 x 2.2	
	<p>(ii) Points plotted correctly ✓ Line of best fit drawn ✓</p> 	2	2 x 2.2	<p>Must have all point within ½ small square IGNORE extrapolation beyond 80°C</p>

Question			Answer	Marks	AO element	Guidance
	(b)	(i)	<p>FIRST CHECK GRAPH FOR CANDIDATE EXPECTED VALUE</p> <p>Reads value from the graph ✓</p> <p>Substitute value into the equation and evaluate $V_{\text{out}} = (10 \div (\text{value from graph} + 10)) \times 10 = \text{evaluated correctly}$ ✓</p> <p>Convert to 1 decimal place ✓</p>	3	<p>2 x 2.2</p> <p>1.2</p>	<p>ALLOW ECF from their graph</p> <p>ALLOW ± ½ small square (0.25) for reading from graph.</p> <p>Example calculation $V_{\text{out}} = (10 \div (14.5 + 10)) \times 10 = 4.08$</p>
		(ii)	Voltage increases ✓	1	3.2b	
		(iii)	<p>FIRST CHECK THE ANSWER ON ANSWER LINE If answer = 0.00077 (A) award 4 marks</p> <p>Total resistance = $R_1 + R_2 = 3000 + 10\,000 = 13\,000$ ✓</p> <p>Recall and rearrange $I = V \div R$ ✓</p> <p>= $10 \div 13000$ ✓</p> <p>= 0.00077 (A) ✓</p>	4	<p>2.2</p> <p>1.2</p> <p>2 x 2.1</p>	<p>ALLOW 7.7×10^{-4} or 77mA for 4 marks</p>

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