



Mark Scheme (Results)

Summer 2012

International GCSE

Chemistry (4CH0) Paper 2C

Edexcel Level 1/Level 2 Certificate

Chemistry (KCH0) Paper 2C

## **Edexcel and BTEC Qualifications**

Edexcel and BTEC qualifications come from Pearson, the world's leading learning company. We provide a wide range of qualifications including academic, vocational, occupational and specific programmes for employers. For further information, please visit our website at [www.edexcel.com](http://www.edexcel.com).

Our website subject pages hold useful resources, support material and live feeds from our subject advisors giving you access to a portal of information. If you have any subject specific questions about this specification that require the help of a subject specialist, you may find our Ask The Expert email service helpful.

[www.edexcel.com/contactus](http://www.edexcel.com/contactus)

## **Pearson: helping people progress, everywhere**

Our aim is to help everyone progress in their lives through education. We believe in every kind of learning, for all kinds of people, wherever they are in the world. We've been involved in education for over 150 years, and by working across 70 countries, in 100 languages, we have built an international reputation for our commitment to high standards and raising achievement through innovation in education. Find out more about how we can help you and your students at: [www.pearson.com/uk](http://www.pearson.com/uk)

Summer 2012

Publications Code UG031850

All the material in this publication is copyright

© Pearson Education Ltd 2012

INTERNATIONAL GCSE CHEMISTRY PAPER 2C – SUMMER 2012

Question number				Answer	Notes	Marks
1	a	(i)	M 1	nitrogen		1
		(ii)	M 1	cross in 2nd box (20)		1
		(iii)	M 1	unreactive		1
			M 2	water		1
	b	(i)	M 1	sulfurous acid / H <sub>2</sub> SO <sub>3</sub>	Accept sulfuric(IV) acid Accept ph spellings If name and formula given, both must be correct	1
		(ii)	M 1	nitrogen oxide(s) / nitrogen dioxide / NO <sub>2</sub> / NO <sub>x</sub>	Ignore nitrogen monoxide / nitrous oxide and other acidic gases, eg carbon dioxide, sulfur trioxide, hydrogen chloride	1

Question number			Answer	Notes	Marks
1	b	(iii)	M 1 iron/steel/metal corrodes/rusts/reacts	Ignore physical process such as erosion/weathering/wearing away/dissolving Ignore burns/burning	2
			M 2 limestone/marble reacts/corrodes/is eaten away NOT just buildings	Ignore rusts or physical process such as erosion/weathering/wearing away/dissolving Ignore burns/burning	
			M 3 plants/trees/vegetation/crops/named example adversely affected in specific way, eg dies/stunted growth/harmed/damaged	Ignore deforestation Ignore leaching minerals	
			M 4 fish/aquatic animals/pond life/marine life/named example affected in specific way, eg dies/stunted growth/harmed/damaged	Ignore references to just animals	
				Accept destroys as an adverse effect for all marks	
				Any two for 1 each	
					<b>Total 8 marks</b>

Question number			Answer	Notes	Marks
2	a	(i)	M1 Iron(III) oxide	Accept Iron oxide / ferric oxide Ignore formula whether right or wrong	1
		(ii)	M1 calcium carbonate	Ignore formula whether right or wrong	1
	b	(i)	M1 A		1
		(ii)	M1 E		1
		(iii)	M1 B		1
		(iv)	M1 C		1
	c		M1 slag	Accept calcium silicate Ignore formula	1
	d	(i)	M1 aluminium/it is more reactive than iron/carbon OR above iron/carbon in reactivity series OR cannot be reduced by/does not react with carbon (monoxide) OR cannot be displaced by carbon	Comparison with iron or carbon must be stated or implied, eg not just aluminium is (very/too) reactive  Accept reverse argument for iron	1
		(ii)	M1 (cost of) electricity	Accept keeping electrolyte molten Accept high current Ignore energy Ignore references to electrode replacement	1

Question number			Answer	Notes	Marks	
2	e		M1	electrode(s) / to conduct electricity	Accept cathode / anode	1
	f		M1	$\text{Al}^{3+} + 3\text{e} \rightarrow \text{Al}$	M1 for both aluminium formulae on correct sides of equation M2 for both oxygen formulae on correct sides of equation M3 for balancing both equations even if one or both reversed	3
			M2	$2\text{O}^{2-} \rightarrow \text{O}_2 + 4\text{e} / 2\text{O}^{2-} - 4\text{e} \rightarrow \text{O}_2$		
			M3			
					Accept in either order	
					<b>Total 13 marks</b>	

Question number			Answer	Notes	Marks
3	a	(i)	M1 more accurate	Accept more precise / gives an exact value	1
		(ii)	M1 (thermal) insulator /poor conductor/keeps heat in/ reduces heat loss	Accepts traps heat	1
		(iii)	M1 stirring/mixing/swirling	Ignore name of apparatus used	1
		(iv)	M1 temperature goes down/stops rising/stays constant	Accept measure pH/when pH = 7/when pH is less than 7 Reject changing to any pH value > 7 Accept use of any indicator (named example or just indicator) Ignore colour changes	1
	b		M1 19.4		1
			M2 23.1		1
			M3 3.7	CQ on temperatures recorded Penalise negative sign	1
				Penalise second decimal place values, except zeroes, for M1 and M2 M1 and M2 both correct but wrong way around scores 1	

Question number			Answer	Notes	Marks
3	c	(i)	M1 9 points plotted correctly to nearest gridline	Deduct 1 for each error	2
			M2	If points not visible beneath line, assume them to be on line	
			M3 <u>straight</u> line of best fit through first 5 points		1
			M4 <u>straight</u> line of best fit through last 4 points		1
				If lines do not cross or are joined by curve or straight line, only M3 or M4 can be awarded	
		(ii)	M1 volume of acid CQ on where lines cross	Accept answer $\pm 0.5$ to minimum of 1 dp	1
			M2 maximum temperature CQ on where lines cross	Accept answer $\pm 0.1$ to minimum of 1 dp	1
				Do not award either mark if lines do not intersect Apply dp penalty once only Award 1 mark if both values correct but recorded in wrong places	
	d		M1 55 ( $\times 4.2 \times 5.5$ )		1
			M2 1270.5	Accept any value between 1270 and 1271 Accept 1300 M2 CQ on M1 If vol used is 25, answer is 577.5 Accept any value between 577 and 578 accept 580 If vol used is 30, answer is 693 Accept 690 Ignore signs	1
				Correct final answer without working scores 2 marks	



Question number			Answer		Notes	Marks
3	e	(i)	M1	$1.5 \times 0.025$		1
			M2	0.0375	Correct final answer scores 2 marks 37.5 scores M2 only	1
		(ii)	M1	$1800 \div 0.0375$ / $1800 \div$ answer to (e)(i)	Accept correct use of 1.8 in place of 1800	1
			M2	48 (kJ/mol)	M2 CQ on M1 provided 1800 or 1.8 used correctly If 37.5 in (a)(i) then answer is 0.048 (kJ/mol) Correct final answer scores 2 marks Ignore answer in J/mol Ignore signs	1
					<b>Total 19 marks</b>	

Question number		Answer	Notes	Marks
4	a	(i) M1 (cane) sugar/ sugar beet	Do not accept any named sugar	1
		(ii) M1 (conc / syrupy) phosphoric acid / H <sub>3</sub> PO <sub>4</sub>	Accept (conc) sulfuric acid / H <sub>2</sub> SO <sub>4</sub> If both name and formula given, both must be correct Ignore oxidation numbers Penalise dilute acid Accept wrong spellings, eg phosphuric acid / phospheric acid	1
		(iii) M1 (to prevent) zymase/enzyme being denatured	Accept zymase/enzyme work best at these temperatures / this is the optimum temperature for zymase/enzyme Accept yeast/zymase/enzyme killed/destroyed	1
	b	reaction 1 reasons	Accept reaction 2 reasons	
		M1 gives pure(r) product / higher (percentage) yield / ethanol is the only product	gives impure product / lower (percentage) yield OR contains solid/sediment/yeast/water Ignore more ethanol produced	3
		M2 reaction fast(er)	reaction 2 slow(er)	
		M3 ethene/C <sub>2</sub> H <sub>4</sub> available (from oil refinery)		
		M4	needs land to grow sugar (cane)	
		M5	needs warm climate	
		M6 reaction continuous	reaction batch	
			Any three for 1 each Ignore references to cost / efficiency	

Question number			Answer	Notes	Marks
4	c		M1 $\text{C}_2\text{H}_5\text{OH} \rightarrow \text{C}_2\text{H}_4 + \text{H}_2\text{O}$	Equation must be balanced Accept reversible arrow Accept structural/displayed formulae Accept word equation	1
			M2 dehydration	Accept elimination / (thermal) decomposition Treat other reaction types (eg reduction) as contradictions	1
					<b>Total 8 marks</b>

Question number			Answer	Notes	Marks	
5	a	(i)	M1	arrow pointing towards negative electrode	Accept by X / on wire / by power supply (as long as pointing in correct direction)	1
		(ii)	M1	hydrogen / H <sub>2</sub>	Ignore H	1
		(iii)	M1	$4\text{OH}^- \rightarrow 2\text{H}_2\text{O} + (1)\text{O}_2 + 4\text{e}^-$	Accept fractions and multiples Accept e in place of e <sup>-</sup> Accept equation with - 4e <sup>-</sup> on LHS	1
	b	(i)	M1	$18 \div 24000$	If division by 24 in place of 24000, no M1 but award M2 for 0.75 No marks for any calculation involving 35.5 or 71 Correct final answer scores 2 marks	1
			M2	$0.00075 / 7.5 \times 10^{-4}$		1
		(ii)	M1	$(b)(i) \times 96500 \times 2$	CQ on (b)(i)	1
			M2	Answer in range 140 - 145 using 0.00075	Correct final answer scores 2 marks Accept answer in range 70 - 72.4 for 1 out of 2 No marks if no use of 96500 or no use of answer from (b)(i)	1

Question number			Answer	Notes	Marks	
5	c	(i)	M1	bromine / Br / Br <sub>2</sub>	Reject bromide / Br <sup>-</sup>	1
		(ii)	M1	reduction <u>and</u> oxidation (at the same time)	Accept oxidisation Ignore oxygenation  Accept loss <u>and</u> gain of electron(s)  Reject loss of electrons by chlorine (molecules) / gain of electrons by bromide (ions)  Reject reduction is loss of electrons / oxidation is gain of electrons  Ignore references to other reaction types, eg displacement / reversible  Ignore references to atoms / ions / molecules / elements	1

Question number			Answer	Notes	Marks	
5	d	(i)	M1	reversible / can go in both directions / (both) forward and reverse reactions can occur	Accept just reference to reverse direction, eg reaction goes backwards / reaction goes in opposite direction Ignore equilibrium	1
		(ii)	M1	shifts to right / moves in forward direction / favours forward reaction/direction	Accept more $\text{PCl}_5$ / product (formed)  Ignore references to rates  M1 can be awarded in explanation part	1
			M2	fewer moles/molecules (of gas) on right / more moles/molecules (of gas) on left / 2 moles/molecules on left and 1 on right / favours side with fewer moles/molecules	Accept particles, but not atoms, in place of molecules  Ignore references to pressure, volume and le Chatelier's principle  Do not award M2 if M1 if shift is to left or no change	1
					<b>Total 12 marks</b>	

PAPER TOTAL: 60 MARKS



Further copies of this publication are available from  
Edexcel Publications, Adamsway, Mansfield, Notts, NG18 4FN

Telephone 01623 467467

Fax 01623 450481

Email [publication.orders@edexcel.com](mailto:publication.orders@edexcel.com)

Order Code xxxxxxxx Summer 2012

For more information on Edexcel qualifications, please visit our website  
[www.edexcel.com](http://www.edexcel.com)

Pearson Education Limited. Registered company number 872828  
with its registered office at Edinburgh Gate, Harlow, Essex CM20 2JE

Ofqual  
■■■■■■■■■■



Llywodraeth Cynulliad Cymru  
Welsh Assembly Government

